

Stoichiometry Worksheet # 1

1) Glucose is used as a source of energy by the human body. The overall reaction in the body is



Calculate the number of moles of oxygen needed to oxidize 12.5 g of $\text{C}_6\text{H}_{12}\text{O}_6$.

2) Ammonia is synthesized from hydrogen and nitrogen according to the following equation.



If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH_3) can be produced?

3) Assume that in the decomposition of potassium chlorate, KClO_3 ,



80.5 moles of O_2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,



9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" $\text{PtCl}_2(\text{NH}_3)_2$, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction



A. How many moles of "cisplatin" can be produced from 2.50 g K_2PtCl_4 ?

B. How many moles of NH_3 would be needed if 3.8 moles of K_2PtCl_4 were present?

Chapter 12 Stoichiometry Test Answer Key

David Kirk



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McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best This book includes 4 full length sample tests updated for the latest test format 40 top tips to remember on test day Glossary of tested chemistry terms and formulas Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School Diagnostic test to pinpoint strengths and weaknesses Step by step review of all topics covered on the exam In depth coverage of the lab experiment questions that are a major test feature Charts tables and illustrations to simplify and reinforce learning Practice tests just like the real SAT Subject Test in Chemistry Test taking tips and strategies

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A First Course in Systems Biology Eberhard O. Voit, 2012-03-28 A First Course in Systems Biology is a textbook designed for advanced undergraduate and graduate students Its main focus is the development of computational models and their applications to diverse biological systems Because the biological sciences have become so complex that no individual can acquire complete knowledge in any given area of specialization the education of future systems biologists must instead develop a student's ability to retrieve reformat merge and interpret complex biological information This book provides the reader with the background and mastery of methods to execute standard systems biology tasks understand the modern literature and launch into specialized courses or projects that address biological questions using theoretical and computational means The format is a combination of instructional text and references to primary literature complemented by sets of small scale exercises that enable hands on experience and larger scale often open ended questions for further reflection

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Kaplan SAT II: Chemistry 2004-2005 Kaplan,2004-03-02 At head of title Kaplan *SAT II* Claire Aldridge,Karla Lee,Kaplan, Inc,2000 Expert review prepares test takers for all the chemistry content they need to know with three full length practice tests and detailed explanations Charts

Trophic Ecology James E. Garvey,Matt Whiles,2016-09-19 This book is a bridge between ecological paradigms organismal community approaches to food web dynamics and ecosystem level approaches to production The unification of organismal community and ecosystem approaches in ecology is emerging due to the growing availability of new techniques for assessing trophic interactions and their implications for ecosystems Trophic Ecology is a formal text for both newcomers to the discipline as well as seasoned professionals looking for new ideas and refreshers on old topics A wide range of topics are explained including autotrophy heterotrophy omnivory decomposition foraging behavior and theory trophic cascades bioenergetics and production The audience is upper level undergraduate students and entry level graduate students interested in autecological organismal approaches to ecology community and ecosystem ecology It is also a reference text for instructors teaching upper division courses providing examples from the literature quantitative approaches to teach and new hypotheses yet to be fully tested by ecologists

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Chemistry Charles H. Corwin,1994 The book focuses on the concepts of chemistry and the applications that maintain and generate motivation for the subject of chemistry

Student Study Guide for Introduction to Chemistry John Dickson,1979-01-25 *Chemical Principles* Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses *Chemical Principles* helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of *Chemical Principles* is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding

Food Analysis Laboratory Manual S. Suzanne Nielsen,2010-03-20 This second edition laboratory

manual was written to accompany Food Analysis Fourth Edition ISBN 978 1 4419 1477 4 by the same author The 21 laboratory exercises in the manual cover 20 of the 32 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component of characteristic Most of the laboratory exercises include the following introduction reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis

Chemistry 2007 David Wilson, 2005-12 Focussing on the material that frequently appears on the AP Chemistry Exam this title includes practice tests with answer explanations diagnostic test practice questions proven strategies and a review of the relevant material

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Water Engineering Nazih K. Shammam, Lawrence K. Wang, 2015-05-27 Details the design and process of water supply systems tracing the progression from source to sink Organized and logical flow tracing the connections in the water supply system from the water's source to its eventual use Emphasized coverage of water supply infrastructure and the design of water treatment processes Inclusion of fundamentals and practical examples so as to connect theory with the realities of design Provision of useful reference for practicing engineers who require a more in depth coverage higher level students studying drinking water systems as well as students in preparation for the FE PE examinations Inclusion of examples and homework questions in both SI and US units

McGraw-Hill's New MCAT George J. Hademenos, 2007 This title is more focused and student friendly than competing MCAT guides It has more special features including a companion CD with two full length timed practice MCATs and free website support There is more bang for the buck this is a complete all in one preparation package for the 60 000 students who take the MCAT every year

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is the only force acting on the sled in x direction and opposes the motion of the sled so the net force is given by die kinetiese wrywingskrag is die enigste krag op die skild in x rigting en

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