Name _____   Date _____   Class _____

Types of Matter

**Use with Chapter 3,
Section 3.4**

1. Into what two broad classes can all matter be divided?

   _____

2. What is the difference between a mixture and a pure substance?

   _____

   _____

3. What is the difference between a compound and an element?

   _____

   _____

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the
   diagram.

   _____

   _____

   _____

5. A list of compounds and elements is given below. Circle the substances that are elements.

gold	aluminum	water
sugar	oxygen	platinum
salt	chlorine	brass

6. How can you tell the difference between a homogeneous mixture and a heterogeneous
   mixture?

   _____

   _____

   _____

7. Label each mixture below as either homogeneous or heterogeneous.

   a. air _____   e. finger paint _____
   b. clay _____   f. vinegar _____
   c. homemade lemonade (with pulp) _____
   d. oatmeal raisin cookie _____   g. soil _____

8. List three methods that are commonly used to separate mixtures into their component
   substances.

   _____

Teaching Transparency Master Answer Key

RS Peters

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Table of Contents Teaching Transparency Master Answer Key

1. Understanding the eBook Teaching Transparency Master Answer Key
    - The Rise of Digital Reading Teaching Transparency Master Answer Key
    - Advantages of eBooks Over Traditional Books
2. Identifying Teaching Transparency Master Answer Key
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Teaching Transparency Master Answer Key
    - User-Friendly Interface
4. Exploring eBook Recommendations from Teaching Transparency Master Answer Key
    - Personalized Recommendations
    - Teaching Transparency Master Answer Key User Reviews and Ratings
    - Teaching Transparency Master Answer Key and Bestseller Lists

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