

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H₂SO₄
 - H₃PO₄
 - HNO₃
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH⁻ ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO₃⁻
 - H₃O⁺
 - H₃PO₄
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- H₂C₂O₄(aq) and CH₃NH₂(aq)
 - H₂C₂O₄(aq) and CH₃NH₃⁺(aq)
 - CH₃NH₂(aq) and CH₃NH₃⁺(aq)
 - HC₂O₄⁻(aq) and CH₃NH₃⁺(aq)

Modern Chemistry Chapter 4 Test Answers

Michal Rosen-Zvi



Modern Chemistry Chapter 4 Test Answers:

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Modern Chemistry Chapter 4 Test Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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