

## Molality Practice

1. Calculate the molarity of each of the following solutions:

a. 12.4 g KCl in 289.2 mL solution

$$M = \frac{n}{V}$$
$$12.4 \text{ g KCl} \times \frac{1 \text{ mole KCl}}{74.6 \text{ g KCl}} = 0.166 \text{ mole KCl}$$

$$289.2 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 0.2892 \text{ L}$$

$$\frac{0.166 \text{ mole}}{0.2892 \text{ L}} = 0.57 \text{ M}$$

## Practice Problem

b. 16.4 g  $\text{CaCl}_2$  in 0.614 L solution

$$M = \frac{n}{V}$$
$$16.4 \text{ g CaCl}_2 \times \frac{1 \text{ mole CaCl}_2}{111.1 \text{ g CaCl}_2} = 0.148 \text{ mole CaCl}_2$$

$$M = \frac{0.148 \text{ mole}}{0.614 \text{ L}} = 0.24 \text{ M}$$

# Molarity Molality Practice Problems Answers

**Patrick Vollmar**



**Molarity Molality Practice Problems Answers:**

## **Molarity Molality Practice Problems Answers** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Molarity Molality Practice Problems Answers**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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