

SOLUTIONS

The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a monobasic base is 1.0×10^{-4} . What is the overall equilibrium constant for the reaction of the fully protonated acid with a dibasic base?

Answer Key 11

1. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a monobasic base is 1.0×10^{-4} . What is the overall equilibrium constant for the reaction of the fully protonated acid with a dibasic base?

2. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .

3. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-4} .

4. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .

5. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-4} .

6. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .

7. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .



8. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .

9. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .



10. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a dibasic base is 1.0×10^{-8} .

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1. The overall equilibrium constant for the reaction of a fully protonated diprotic acid with a monobasic base is 1.0×10^{-4} . What is the overall equilibrium constant for the reaction of the fully protonated acid with a dibasic base?

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