

Lewis Structure Worksheet #2

Key

Draw Lewis Structures for each of the following molecules or ions:

1. Br ₂ 14 $:\ddot{\text{Br}}-\ddot{\text{Br}}:$	2. BH ₃ 6 $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{B}-\text{H} \\ \\ \text{H} \end{array}$
3. O ₂ 12 $:\ddot{\text{O}}=\ddot{\text{O}}:$	4. PS ₃ ⁻ 24 $\begin{array}{c} :\ddot{\text{S}} \\ \\ \ddot{\text{S}}=\text{P}-\ddot{\text{S}}: \\ \\ :\ddot{\text{S}}: \end{array} + \text{resonance}$
5. H ₂ CS 12 $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}=\ddot{\text{S}}: \end{array}$	6. CHCl ₃ 26 $\begin{array}{c} :\ddot{\text{Cl}}: \\ \\ \text{H}-\text{C}-\ddot{\text{Cl}}: \\ \\ :\ddot{\text{Cl}}: \end{array}$
7. SF ₂ 20 $:\ddot{\text{F}}-\ddot{\text{S}}-\ddot{\text{F}}:$	8. SHF 14 $\text{H}-\ddot{\text{S}}-\ddot{\text{F}}:$
9. N ₂ F ₄ 38 $\begin{array}{c} \text{F} \quad \text{F} \\ \diagdown \quad / \\ \text{N}-\text{N} \\ / \quad \diagdown \\ \text{F} \quad \text{F} \end{array}$	10. XeO ₄ 32 $\begin{array}{c} \text{O} \quad \text{O} \\ \diagdown \quad / \\ \text{Xe} \\ / \quad \diagdown \\ \text{O} \quad \text{O} \end{array}$
11. N ₂ O ₃ (dinitrogen trioxide)* 28e $:\ddot{\text{O}}=\ddot{\text{N}}-\text{N}=\ddot{\text{O}}: + \text{resonance}$	12. NO ₂ (nitrogen dioxide)* 17e $:\ddot{\text{O}}=\ddot{\text{N}}-\ddot{\text{O}}: + \text{resonance}$
13. N ₂ O ₄ (dinitrogen tetraoxide)* 34 $\begin{array}{c} \text{O} \quad \text{O} \\ \diagdown \quad / \\ \text{N}-\text{N} \\ / \quad \diagdown \\ \text{O} \quad \text{O} \end{array} + \text{resonance}$	14. N ₂ O ₅ (dinitrogen pentaoxide)* 40e $\begin{array}{c} \text{O} \quad \text{O} \quad \text{O} \\ \diagdown \quad / \quad \diagdown \quad / \\ \text{N}-\text{O}-\text{N} \\ / \quad \diagdown \quad / \quad \diagdown \\ \text{O} \quad \text{O} \quad \text{O} \end{array} + \text{resonance}$

*Problems 11-14 are more challenging and may have more than one correct structure. Please note that if a molecule has an odd number of valence electrons, at least one atom in the molecule will NOT fulfill the octet rule.

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DJ Losen



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