

Name: Xinya ZengChem 10, Section: 1105

Lab Partner: \_\_\_\_\_

Exercise Date: 01/27/21

## Lewis Structures and Molecular Shapes

1.  $\text{H}_2\text{S}$ 

Total # of Valence Electrons: $1 \times 2 + 6 = 8e^-$ (4P)	3-D Model Sketch	Bond Angles
Lewis Structure (show all resonance structures if applicable)		$109.5^\circ$
Any polar bonds in the molecule? <input checked="" type="radio"/> Yes <input type="radio"/> No	Molecular Shape: <u>Bent</u>	Molecular Polarity: <input checked="" type="radio"/> Polar <input type="radio"/> Non-Polar

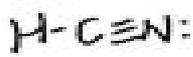
2.  $\text{COCl}_2$ 

Total # of Valence Electrons: $4 + 6 + 2 \times 7 = 24e^-$ (12P)	3-D Model Sketch	Bond Angles
Lewis Structure (show all resonance structures if applicable)		$120^\circ$
Any polar bonds in the molecule? <input checked="" type="radio"/> Yes <input type="radio"/> No	Molecular Shape: <u>Trigonal planar</u>	Molecular Polarity: <input checked="" type="radio"/> Polar <input type="radio"/> Non-Polar

3.  $\text{SiH}_4$  (use a black ball for silicon when constructing this model)

Total # of Valence Electrons: $4 + 4 \times 7 = 32e^-$ (16P)	3-D Model Sketch	Bond Angles
Lewis Structure (show all resonance structures if applicable)		$109.5^\circ$
Any polar bonds in the molecule? <input checked="" type="radio"/> Yes <input type="radio"/> No	Molecular Shape: <u>Tetrahedral</u>	Molecular Polarity: <input type="radio"/> Polar <input checked="" type="radio"/> Non-Polar

4.  $\text{HCN}$ 

Total # of Valence Electrons: $1 + 4 + 5 = 10e^-$ (5P)	3-D Model Sketch	Bond Angles
Lewis Structure (show all resonance structures if applicable)		$180^\circ$
Any polar bonds in the molecule? <input checked="" type="radio"/> Yes <input type="radio"/> No	Molecular Shape: <u>Linear</u>	Molecular Polarity: <input checked="" type="radio"/> Polar <input type="radio"/> Non-Polar

# Lewis Structures Molecular Model Lab Answers

**JR Anderson**



## **Lewis Structures Molecular Model Lab Answers:**

Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives *Essentials of General Organic and Biochemistry* captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit [www.whfreeman.com/gob](http://www.whfreeman.com/gob)

**Inquiries into Chemistry** Michael R. Abraham, Michael J. Pavelich, 1999-05-20 The laboratory course should do more than just acquaint the students with fundamental techniques and procedures The laboratory experience should also involve the students in some of the kinds of mental activities a research scientist employs finding patterns in data developing mathematical analyses for them forming hypotheses testing hypotheses debating with colleagues and designing experiments to prove a point For this reason the student tested lab activities in *Inquiries into Chemistry 3 E* have been designed so that students can practice these mental activities while building knowledge of the specific subject area Instructors will enjoy the flexibility this text affords They can select from a comprehensive collection of structured guided inquiry experiments and a corresponding collection of open inquiry experiments depending on their perception as to what would be the most appropriate method of instruction for their students Both approaches were developed to encourage students to think logically and independently to refine their mental models and to allow students to have an experience that more closely reflects what occurs in actual scientific research Thoroughly illustrated appendices cover safety in the lab common equipment and procedures

*Cooperative Chemistry Lab Manual* Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations

Conformational Concept For Synthetic Chemist's Use: Principles And In Lab Exploitation Anatoly M Belostotskii, 2015-09-17 This innovative book presents an original account of the principles of conformational theory It has a strong focus on computational methodologies for conformational space exploration By revisiting basic conformational conventions considering experimental results which are often misinterpreted by organic chemists and qualitatively analyzing the potential energy surface the book helps non experts to understand molecular flexibility at the level required in contemporary research The book shows synthetic organic chemists

how to perform successful conformational studies using widespread calculation packages click computational chemistry instead of being misguided by textbook based conformational analysis The monograph actually offers to synthetic chemists a new research tool that can significantly upgrade their ability to predict or at least explain regioselectivity and stereoselectivity in their own reactions *Exploring General Chemistry in the Laboratory* Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Experiments and Exercises in Basic Chemistry Steven Murov, Brian Stedjee, 2003-03-12 Taking an exploratory approach to chemistry this hands on lab manual for preparatory chemistry encourages critical thinking and allows students to make discoveries as they experiment A set of exercises provides students with additional opportunities to test their understanding of key concepts in introductory and prep chemistry courses Written in a clear easy to read style Numerous experiments to choose from cover all topics typically covered in prep chemistry courses Chemical Capsules demonstrate the relevance and importance of chemistry Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **Instructors Manual to Lab Manual** Ralph Petrucci, William Harwood, Geoffrey Herring, 2001

**Laboratory Manual for Principles of General Chemistry** J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write

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**Chemistry** McGraw-Hill Staff, 2001-07 **AP Chemistry** Albert S. Tarendash, Frederick J. Rowe, 2002 This indispensable guide to chemistry helps students who wish to prepare for the AP Chemistry exam on their own Comprehensive and easy to understand this learning guide includes a full content review two full length practice tests with hundreds of practice questions and thorough answer explanations and proven test taking strategies Polarity, Solutions, and Separation Science Kenda Jo Lemont, 1998 **Introduction to General, Organic, and Biochemistry** Morris Hein, 1997 The authors recognize that both science and mathematics may be daunting subjects for many students taking this course With this in mind they have anticipated where students might stumble and have paced and organized this text to help them through Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career The authors emphasize problem solving and provide a range of practice exercises As in previous editions the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry In this edition the first two sections have been revised primarily to improve explanations and include new pedagogical features The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field

**The Software Encyclopedia 2000** Bowker Editorial Staff, 2000-05 **Exploring Chemistry Laboratory Experiments in General, Organic and Biological Chemistry** Julie R. Peller, 2003-04 This lab manual is organized and written to ensure that non science majors are comfortable with chemistry labs by making the experiments more applicable to students' daily lives This approach also serves to make the experiments more understandable Many labs relate specifically to allied health fields Chemistry Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives

**Laboratory Manual for Fundamentals of Chemistry 3/E** Jo A. Beran, 1988

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