

CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE & PERIODIC TRENDS- KEY

1) When a solution of sodium chloride is vaporized in a flame, the color of the flame is:

- A) Blue **B) Yellow** C) Green D) Violet E) White

2) Electromagnetic radiation travels through vacuum at a speed of _____ m/s.

- A) 186,000 **C) 3.00×10^8** E) It depends on λ
 B) 125 D) 10,000

3) The wavelength of light that has a frequency of $1.20 \times 10^{13} \text{ s}^{-1}$ is _____ m.

- A) 25.0 **B) 2.50×10^{-2}** C) 0.0400 E) 2.5
 D) 12.0

GIVEN: $\nu = 1.20 \times 10^{13} \text{ s}^{-1}$

UNKNOWN: λ

CONNECTION: $c = \lambda \nu$

SUBSTITUTE & SOLVE: $\lambda = c/\nu$

$$\lambda = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{1.20 \times 10^{13} \text{ s}^{-1}}$$

$$\lambda = 2.50 \times 10^{-2} \text{ m}$$

4) Ham radio operators often broadcast on the 6-meter band. The frequency of this EM radiation is _____ MHz.

- A) 500 B) 200 **C) 50** D) 20 E) 2.0

GIVEN: $\lambda = 6 \text{ m}$

UNKNOWN: ν (Must convert answer to MHz!)

CONNECTION: $c = \lambda \nu$

SUBSTITUTE & SOLVE: $\nu = c/\lambda$

$$\nu = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{6 \text{ m}}$$

$$\nu = 0.5 \times 10^8 \text{ Hz} \times \frac{1 \text{ MHz}}{10^6 \text{ Hz}} = 50 \text{ MHz}$$

5) The energy of a photon of light is _____ proportional to its frequency and _____ proportional to its wavelength.

- A) directly, directly C) inversely, directly E) indirectly, not
 B) inversely, inversely **D) directly, inversely**

6) The energy of a photon that has a wavelength of 12.3 nm is _____ J.

- A) 1.51×10^{-17} B) 4.42×10^{-23} C) 1.99×10^{-23} D) 2.72×10^{-20} **E) 1.62×10^{-17}**

7) The wavelength of a photon that has an energy of $5.25 \times 10^{-19} \text{ J}$ is _____ m.

- A) 3.79×10^{-7}** B) 2.64×10^6 C) 2.38×10^{23} D) 4.21×10^{-24} E) 3.79×10^7

8) A mole of red photons of wavelength 725 nm has _____ kJ of energy.

- A) 2.74×10^{-29} B) 4.56×10^{-26} C) 6.05×10^{-7} **D) 165** E) 227

GIVENS: $\lambda = 725 \text{ nm}$ (must convert to m before using formulas)

$$\lambda = 725 \text{ nm} \times \frac{1 \text{ m}}{10^9 \text{ nm}} = 7.25 \times 10^{-7} \text{ m}$$

Holt Chemistry Chapter 7 Test

Paul R. Loconto



Holt Chemistry Chapter 7 Test:

Holt Chemistry Ralph Thomas Myers,2004 **Books in Print Supplement** ,2002 **Holt Chemistry File** ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket *Children's Books in Print, 2007* ,2006 *El-Hi Textbooks and Serials in Print, 1996* R R Bowker Publishing,Bowker,1996-02 **Trace**

Environmental Quantitative Analysis Paul R. Loconto,2020-12-27 A thorough and timely update this new edition presents principles techniques and applications in this sub discipline of analytical chemistry for quantifying traces of potentially toxic organic and inorganic chemical substances found in air soil fish and water as well as serum plasma urine and other body fluids The author addresses regulatory aspects calibration verification and the statistical treatment of analytical data including instrument detection limits quality assurance quality control sampling and sample preparation and techniques that are used to quantify trace concentrations of organic and inorganic chemical substances Key Features Fundamental principles are introduced for the more significant experimental approaches to sample preparation Principles of instrumental analysis determinative techniques for trace organics and trace inorganics analysis An introduction to the statistical treatment of trace analytical data How to calculate instrument detection limits based on weighted least squares confidence band calibration statistics Includes an updated series of student tested experiments Holt Biology Holt Rinehart & Winston,2003-08

Children's Books in Print R R Bowker Publishing,Bowker,1999-12 **Holt Algebra 1 2003** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003 El-Hi Textbooks & Serials in Print, 2005 ,2005 *Keys to ... Tests in Chemistry Series A and Series B* Charles Elwood Dull,1958 Construction of a Diagnostic Test in the Mechanics and Related Fundamentals of High School Chemistry ... Joseph Earl Malin,1932 **Instructional Patterns** Larry C. Holt,Marcella L. Kysilka,2006 I like the mix of theory and research background with thorough descriptions of classroom use vignettes and how to s Teresa Secules Piedmont College Instructional Patterns Strategies for Maximizing Student Learning examines instruction from the learners point of view by showing how instructional patterns can be used to maximize the potential for students to learn This book explores the interactive patterns that exist in today s classroom and demonstrates how teachers can facilitate the interactivity of these patterns to match their goals for student learning These interactive patterns are reinforced through the incorporation of medical cognitive and behavioral neuroscience research This unique book will serve as a core text for undergraduate and graduate courses in K 12 General Teaching Methods Middle School and Secondary Teaching Methods Elementary Teaching Methods or Instruction and Assessment Key Features Guides students in differentiating instructional practices to meet the needs of all students as well as in the practical issues of instruction Details interactive instructional patterns that include teacher centered patterns teacher student interactive patterns and student

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