

ACS BIOCHEMISTRY EXAM QUESTIONS AND ANSWERS

Chemical Equilibrium Equations - Answer: $\mu_A + \mu_B = \mu_C + \mu_D$ (1/100)

Free Energy Change - Answer: Total free energy of a reacting system does not change in a reversible process. $\Delta G_{total} = 0$ (1/100)

Le Chatelier's Principle - Answer: Changes in conditions such as temperature, pressure, concentration affect the changes in the equilibrium for changes in the system.

Electrochemical Cells - Answer: Separates sample (analyte) into two electrodes (anode and cathode).

Electrochemical Cells - Answer: Separates sample (analyte) into two electrodes (anode and cathode). The cell potential is measured across the electrodes. The cell potential is related to the concentration of the sample.

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