

Name Key

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CHAPTER 10 STUDY GUIDE FOR CONTENT MASTERY

Chemical Reactions

Section 10.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- _____ yes _____ 1. A tomato smells rotten.
- _____ no _____ 2. A drinking glass breaks into smaller pieces.
- _____ no _____ 3. A piece of ice melts.
- _____ yes _____ 4. Drain cleaner is mixed with water and the solution becomes warm.
- _____ yes _____ 5. Candle wax burns.
- _____ no _____ 6. Molten candle wax solidifies.
- _____ yes _____ 7. Green leaves turn yellow and red as the seasons change.
- _____ yes _____ 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) (9) reactant, oxygen is a(n) (10) reactant, and water vapor is a(n) (11) product. In a chemical equation for this reaction, a(n) (12) arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) (13) plus sign is used to separate the symbols for hydrogen and oxygen. A(n) (14) (g) symbol is used to tell the state of hydrogen in the reaction, a(n) (15) (g) symbol is used for the state of oxygen, and a(n) (16) (g) symbol is used for the state of water vapor.

Modern Chemistry Chapter 3 Test Answer Key

**Donald Voet, Judith G. Voet, Charlotte
W. Pratt**



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