



Determination Of % Available Chlorine In Bleaching Powder

BY
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17



Aim

To determine the available chlorine in the given sample of bleaching powder by the iodometric method.



Introduction

- ❖ Bleaching powder, also known as chlorinated lime
- ❖ yellowish-white powder having a smell of chlorine and is readily soluble in water
- ❖ It is prepared by passing chlorine gas over slaked lime at a temperature of 35-45° and consists of a mixture of calcium hypochlorite $\text{Ca}(\text{OCl}_2)$ and calcium chloride CaCl_2 , in addition some amount of free slaked lime i.e. $\text{Ca}(\text{OH})_2 \cdot \text{H}_2\text{O}$ is generally present
- ❖ $\text{Ca}(\text{OCl}_2)$ is responsible for the bleaching action of bleaching powder. On treatment with glacial acetic acid, it liberates chlorine gas (Cl_2) as per the following reaction.



- ❖ The amount of chlorine liberated by the action of an acid on bleaching powder (CaOCl_2) is termed as available chlorine. The chlorine content of bleaching powder varies from 35 – 40%.



- ❖ bleaching powder also have strong germicidal and disinfectant properties.
- ❖ It finds application as a disinfectant for drinking water or swimming pool water.

Determination Of Available Chlorine In Bleaching Solution

Clarence Jay West



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Table of Contents Determination Of Available Chlorine In Bleaching Solution

1. Understanding the eBook Determination Of Available Chlorine In Bleaching Solution
 - The Rise of Digital Reading Determination Of Available Chlorine In Bleaching Solution
 - Advantages of eBooks Over Traditional Books
2. Identifying Determination Of Available Chlorine In Bleaching Solution
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Determination Of Available Chlorine In Bleaching Solution
 - User-Friendly Interface
4. Exploring eBook Recommendations from Determination Of Available Chlorine In Bleaching Solution
 - Personalized Recommendations
 - Determination Of Available Chlorine In Bleaching Solution User Reviews and Ratings
 - Determination Of Available Chlorine In Bleaching Solution and Bestseller Lists
5. Accessing Determination Of Available Chlorine In Bleaching Solution Free and Paid eBooks

- Determination Of Available Chlorine In Bleaching Solution Public Domain eBooks
 - Determination Of Available Chlorine In Bleaching Solution eBook Subscription Services
 - Determination Of Available Chlorine In Bleaching Solution Budget-Friendly Options
6. Navigating Determination Of Available Chlorine In Bleaching Solution eBook Formats
- ePub, PDF, MOBI, and More
 - Determination Of Available Chlorine In Bleaching Solution Compatibility with Devices
 - Determination Of Available Chlorine In Bleaching Solution Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Determination Of Available Chlorine In Bleaching Solution
 - Highlighting and Note-Taking Determination Of Available Chlorine In Bleaching Solution
 - Interactive Elements Determination Of Available Chlorine In Bleaching Solution
8. Staying Engaged with Determination Of Available Chlorine In Bleaching Solution
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Determination Of Available Chlorine In Bleaching Solution
9. Balancing eBooks and Physical Books Determination Of Available Chlorine In Bleaching Solution
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Determination Of Available Chlorine In Bleaching Solution
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Determination Of Available Chlorine In Bleaching Solution
- Setting Reading Goals Determination Of Available Chlorine In Bleaching Solution
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Determination Of Available Chlorine In Bleaching Solution
- Fact-Checking eBook Content of Determination Of Available Chlorine In Bleaching Solution
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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