

**123**

123-446452

Regd.

**I**

Total No. of Questions - 21

Total No. of Printed Pages - 2

No.

**Part - III**  
**CHEMISTRY - Paper - I**  
**(English Version)**

**Time : 3 Hours****Max. Marks : 60**

**Note :- Read the following instructions carefully.**

- (i) Answer all questions of Section-A. Answer any six questions out of eight questions from Section-B and answer any two questions out of three questions from Section-C.
- (ii) In Section-A, questions from Sr. Nos. 1 to 10 are of "Very Short Answer Type". Each question carries two marks. Every answer may be limited to two or three sentences. Answer all these at one place in the same order.
- (iii) In Section-B, questions from Sr. Nos. 11 to 18 are of "Short Answer Type". Each question carries four marks. Every answer may be limited to 75 words.
- (iv) In Section-C, questions from Sr. Nos. 19 to 21 are of "Long Answer Type". Each question carries eight marks. Every answer may be limited to 300 words.
- (v) Draw labelled diagrams wherever necessary for questions in Section-B and Section-C.

**SECTION - A**

**10×2=20**

**Note :- Answer all questions :**

1. Lithium salts are mostly hydrated. Why?
2. What is Chemical Oxygen Demand (COD)?
3. Which oxides cause acid rain and what is its pH value?
4. State Dalton's Law of Partial Pressure.
5. The empirical formula of a compound is  $\text{CH}_2\text{O}$ . Its molecular weight is 90. Calculate the molecular formula of the compound.
6. What is meant by ionic product of water?
7. What happens when magnesium metal is burnt in air?



# Diploma Polytechnic 1st Year Chemistry Question Paper

**Léo Pomar**



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