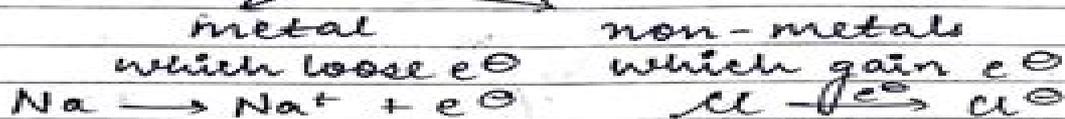


# Periodic Table

Page No. 75

Lavoisier classification :-



cannot classify metalloids.

Prout's Hypothesis

All elements are made up of hydrogen.

$$At. wt. = n \times (At. wt. of H atom)$$

$$n = 1, 2, 3, 4, \dots$$

Drawback :-  $Cl \rightarrow 35.5$  it is not whole no.

Dobereiner law of triads :-

3 elements  $\rightarrow$  Triad

[1, 2, 3]

$$w_2 = \frac{w_1 + w_3}{2}$$

Drawback: A large no. of elements can't be classified in triads.

eg	Li	7	$\frac{7 + 39}{2} = 23$
	Na	23	
	K	39	
eg	Cl	35.5	$\frac{35.5 + 127}{2} = 81.5$
	Br	80	
	I	127	

Fe + Co Ni X

55 58.9 58.6

# Inorganic Chemistry Notes For Iit Jee

**AW Chickering**



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