

7.3

BONDING IN METALS

Section Review

Objectives

- 1. Describe metallic bonding
- 2. Describe the delocalized electron model
- 3. Explain the properties of metals

Vocabulary

- metallic bond
- metal

Part A: Completion

Use the appropriate word or phrase to complete each statement. Write your answer in the space provided.

1. The model of metallic bonding is called the **delocalized electron model**.
2. **Electrons** are delocalized in a metal.
3. **Electrons** are delocalized in a metal.
4. **Electrons** are delocalized in a metal.
5. **Electrons** are delocalized in a metal.
6. **Electrons** are delocalized in a metal.
7. **Electrons** are delocalized in a metal.
8. **Electrons** are delocalized in a metal.
9. **Electrons** are delocalized in a metal.
10. **Electrons** are delocalized in a metal.

11. **Electrons** are delocalized in a metal.
12. **Electrons** are delocalized in a metal.
13. **Electrons** are delocalized in a metal.
14. **Electrons** are delocalized in a metal.
15. **Electrons** are delocalized in a metal.
16. **Electrons** are delocalized in a metal.
17. **Electrons** are delocalized in a metal.
18. **Electrons** are delocalized in a metal.
19. **Electrons** are delocalized in a metal.
20. **Electrons** are delocalized in a metal.

Part B: True/False

Write **True** or **False** in the space provided for each statement.

1. **True** Electrons are delocalized in a metal.
2. **True** Electrons are delocalized in a metal.

Bonding In Metals Section Review Answers Key

Liyong Dong



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