

ICSE Class 10 Chemistry Question Paper Solution 2016

CHEMISTRY SCIENCE Paper – 2

Question 1

- (a) Fill in the blanks with the choices given in brackets. [5]
- (i) Metals are good _____ (*oxidizing agents / reducing agents*) because they are electron _____ (*acceptors/ donors*).
- (ii) Electrovalent compounds have _____ (*high / low*) melting points.
- (iii) Higher the pH value of a solution, the more _____ (*acidic / alkaline*) it is.
- (iv) _____ (*AgCl / PbCl₂*), a white precipitate is soluble in excess NH_4OH .
- (v) Conversion of ethene to ethane is an example of _____ (*hydration / hydrogenation*).
- (b) Choose the *correct answer* from the options given below: [5]
- (i) An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
- A. 17
B. 11
C. 18
D. 20
- (ii) The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
- A. 1:2
B. 1:0.01
C. 1:1
D. 0.01:1 [Given that H=1, O=16]
- (iii) The two main metals in *Bronze* are:
- A. Copper and zinc
B. Copper and lead
C. Copper and nickel
D. Copper and tin
- (iv) The particles present in *strong electrolytes* are:
- A. only molecules
B. mainly ions

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University of the State of New York



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