

Name : _____ Date : _____

Time : Start End

Marks : 280

DPP JEE Main

Class XI | C | 033

CHEMISTRY

Chapter 12 : Organic Chemistry : Some Basic Principles and Techniques

Topics :

- Basics of Organic Chemistry
- Nomenclature of Organic Compounds
- Isomerism-Structural and Stereoisomerism

Instructions :

- ☐ DPP contains 70 objective questions. ☐ Each DPP contains Multiple Choice Questions (MCQs) and Numerical Value Type Questions. ☐ Each question carries 4 marks.
- ☐ Mark the correct answer for MCQs and answers to be filled in as a numerical value for Numerical Value Type Questions in the OMR Sheet given at the end of the DPP. ☐ For every incorrect answer deduct 1 mark.

Basics of Organic Chemistry

1. The number of σ - and π -bonds in 5-oxohexanoic acid, respectively, is
 - (a) 18, 2
 - (b) 18, 1
 - (c) 17, 2
 - (d) 17, 1
2. Which of the following statements is wrong for a homologous series?
 - (a) All members have the same general formula.
 - (b) All members have the same functional group.
 - (c) All members have the same chemical properties.
 - (d) All members have the same physical properties.
3. Which of the following is a 3° alcohol?
 - (a) *t*-Butyl carbinol
 - (b) 2-Methylpropan-2-ol
 - (c) 2-Methylbutan-1-ol
 - (d) Benzyl alcohol
4. The number of σ - and π -bonds in hexane-2,4-dione respectively, is
 - (a) 18, 2
 - (b) 17, 2
 - (c) 17, 1
 - (d) 18, 2
5. The hybridisations of C atoms in (C - C) single-bond of $\text{H}-\text{C}\equiv\text{C}-\text{CH}=\text{CH}_2$ is
 - (a) $sp^2 - sp^2$
 - (b) $sp^2 - sp^3$
 - (c) $sp - sp^2$
 - (d) $sp^2 - sp$
6. Polarisation of electrons in acrolein may be written as:
 - (a) $\overset{\ominus}{\text{C}}\text{H}_2 = \text{CH} - \overset{\oplus}{\text{C}} = \text{O}$
 - (b) $\overset{\oplus}{\text{C}}\text{H}_2 = \text{CH} - \text{CH} = \overset{\ominus}{\text{O}}$
 - (c) $\overset{\oplus}{\text{C}}\text{H}_2 = \overset{\ominus}{\text{C}} - \text{CH} = \text{O}$
 - (d) $\overset{\oplus}{\text{C}}\text{H}_2 = \text{CH} - \text{CH} = \overset{\oplus}{\text{O}}$
7. The bond between carbon atom (1) and carbon atom (2) in the compound $\text{N}\equiv\overset{1}{\text{C}} - \overset{2}{\text{C}}\text{H} = \text{CH}_2$ involves the hybridisation as
 - (a) sp^2 and sp^2
 - (b) sp^3 and sp
 - (c) sp and sp^2
 - (d) sp and sp
8. The number of 1°, 2°, and 3° H atoms in 3-ethyl-5-methyl heptane, respectively, is
 - (a) 12, 8, 1
 - (b) 14, 4, 2
 - (c) 12, 6, 2
 - (d) 12, 8, 2

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