

TEACHING TRANSPARENCY WORKSHEET

Types of Matter

**Use with Chapter 3,
Section 3.4**

1. Into what two broad classes can all matter be divided?

2. What is the difference between a mixture and a pure substance?

3. What is the difference between a compound and an element?

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.

5. A list of compounds and elements is given below. Circle the substances that are elements.

- | | | |
|-------|----------|----------|
| gold | aluminum | water |
| sugar | oxygen | platinum |
| salt | chlorine | brass |

6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?

7. Label each mixture below as either homogeneous or heterogeneous.

- | | |
|--|-----------------------|
| a. air _____ | e. finger paint _____ |
| b. clay _____ | f. vinegar _____ |
| c. homemade lemonade (with pulp) _____ | g. soil _____ |
| d. oatmeal raisin cookie _____ | |

8. List three methods that are commonly used to separate mixtures into their component substances.

Teaching Transparency Master 39 Chemistry Answers

R Pring



Teaching Transparency Master 39 Chemistry Answers:

Unveiling the Magic of Words: A Review of "**Teaching Transparency Master 39 Chemistry Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Teaching Transparency Master 39 Chemistry Answers**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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