

REPORT SHEET  
**Acid-Base Titration**

LAB  
**10**

**A. Concentration of Acetic Acid in Vinegar**

1. Brand Heinz Volume 5.00 mL (% on label) 5.0 %  
 2. Molarity (M) of NaOH \_\_\_\_\_ M

	Trial 1	Trial 2	Trial 3
3. Initial NaOH level in buret	<u>0.1 mL</u>	<u>0.2 mL</u>	<u>0.1 mL</u>
4. Final NaOH level in buret	<u>39.6 mL</u>	<u>41.2 mL</u>	<u>40.1 mL</u>
5. Volume (mL) of NaOH used			
6. Average volume (mL)			
7. Average volume in liters (L)			

Show all calculations for the following questions.

8. Moles of NaOH used in titration \_\_\_\_\_ mole NaOH
9. Moles of  $\text{HC}_2\text{H}_3\text{O}_2$  neutralized by NaOH \_\_\_\_\_ mole  $\text{HC}_2\text{H}_3\text{O}_2$
10. Molarity of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ M  $\text{HC}_2\text{H}_3\text{O}_2$
11. Grams of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ g  $\text{HC}_2\text{H}_3\text{O}_2$
12. Percent (m/v)  $\text{HC}_2\text{H}_3\text{O}_2$  in vinegar \_\_\_\_\_ %  $\text{HC}_2\text{H}_3\text{O}_2$

# Acid Base Titration Lab Answers Ap Chem Parncs

**Princeton Review**



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