

Chapter 4 Atomic Structure

Section 4.2 The Structure of an Atom**(pages 108–112)**

This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.

Reading Strategy (page 108)

Monitoring Your Understanding Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Know About Atoms	What I Would Like to Learn	What I Have Learned
Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

Properties of Subatomic Particles (pages 108–109)

- What are three subatomic particles?
a. Protons b. Electrons c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.
a. nucleus **(b)** proton
c. neutron d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.

Comparing Subatomic Particles (pages 109–110)

- Circle the letters of properties that vary among subatomic particles.
a. color **(b)** mass
(c) charge **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.
(a) mass of 1 neutron = mass of 1 proton
b. mass of 2000 neutrons = mass of 1 proton
c. mass of 1 electron = mass of 1 proton
d. mass of 1 neutron = mass of 1 electron

Chapter 4 Atomic Structure Answers

**Kaplan, Kaplan Educational Center
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Chapter 4 Atomic Structure Answers:

New York State Regents Exam Kaplan, Kaplan Educational Center Staff, 2000-02 Kaplan's guides to the New York State Regents Exams come complete with a comprehensive review of all the tested material plus Kaplan's exclusive test taking strategies. This powerful combination makes the New York State Regents Exam Chemistry Second Edition a highly effective way for you to score higher on this very challenging test. Are you ready for the New York State Regents Chemistry exam? You will be with Kaplan's proven plan for success.

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[The Science For Conservators Series](#) The Conservation Unit, Museums and Galleries Commission, 2008-05-20 For more than ten years, The Science for Conservators Series has provided the key basic texts for conservators throughout the world. Scientific concepts are basic to the conservation of artefacts of every type, yet many conservators have little or no scientific training. These introductory volumes provide non-scientists with the essential theoretical background to their work.

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Study Guide with Answers to Selected Problems H. Stephen Stoker,Danny V. White,2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises **MATERIALS SCIENCE AND ENGINEERING, SIXTH EDITION** V. RAGHAVAN,2015-05-01 This well established and widely adopted book now in its Sixth Edition provides a thorough analysis of the subject in an easy to read style It analyzes systematically and logically the basic concepts and their applications to enable the students to comprehend the subject with ease The book begins with a clear exposition of the background topics in chemical equilibrium kinetics atomic structure and chemical bonding Then follows a detailed discussion on the structure of solids crystal imperfections phase diagrams solid state diffusion and phase transformations This provides a deep insight into the structural control necessary for optimizing the various properties of materials The mechanical properties covered include elastic anelastic and viscoelastic behaviour plastic deformation creep and fracture phenomena The next four chapters are devoted to a detailed description of electrical conduction superconductivity semiconductors and magnetic and dielectric properties The final chapter on Nanomaterials is an important addition to the sixth edition It describes the state of art developments in this new field This eminently readable and student friendly text not only provides a masterly analysis of all the relevant topics but also makes them comprehensible to the students through the skillful use of well drawn diagrams illustrative tables worked out examples and in many other ways The book is primarily intended for undergraduate students of all branches of engineering B E B Tech and postgraduate students of Physics Chemistry and Materials Science **KEY FEATURES** All relevant units and constants listed at the beginning of each chapter A note on SI units and a full table of conversion factors at the beginning A new chapter on Nanomaterials describing the state of art information Examples with solutions and problems with answers About 350 multiple choice questions with answers McGraw-Hill's 500 MCAT General Chemistry Questions to Know by Test Day John T. Moore,Richard H. Langley,2012-06-29 A wealth of problem solving practice in the format that you want This book is the ideal way to sharpen skills and prepare for this MCAT topic Get the problem solving practice for general chemistry you need with McGraw Hill s 500 MCAT General Chemistry

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Chapter 4 Atomic Structure Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chapter 4 Atomic Structure Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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