

The Periodic Law

SECOND PERIODIC

NOTE: All answers should include the following questions in the space provided.

- Classify the central atoms in the following compounds as metals, metalloids, or nonmetals.

Ca	_____	a. Which is the more reactive?
Cl	_____	b. Which is the more electronegative?
Aluminum	_____	c. Is the element in position a or position b more likely to be oxidized by an element in position a or by an element in position b?
- Classify the elements below as metals, metalloids, or nonmetals.

Li	_____	a. For which element does the atomic number Z equal $A - 2$?
Ag	_____	b. Z is an integer which is the value of n ?
Fe	_____	c. The group number is the same as $Z - 2$ or $Z - 10$?
- | | | |
|----|-------|--|
| Br | _____ | a. Is electronegativity generally lower (L) or higher (H) than Z ? |
| Al | _____ | b. Is electronegativity generally lower (L) or higher (H) than Z ? |
- | | | |
|-----------|-------|--|
| Br | _____ | a. Does the more electronegative element have a higher Z ? |
| Li and Cl | _____ | b. Which is the more electronegative? |
| Li and Cl | _____ | c. Which is the more reactive of the more electronegative element? |
- Using only the periodic table, write the following series of elements in order of increasing atomic number.

Ca	_____	a. Which has the largest atomic number?
Cl	_____	b. Which has the largest atomic number?
Al	_____	c. Which has the largest atomic number?
Aluminum	_____	d. Which has the largest atomic number?
Cl	_____	e. Which has the largest atomic number?
Argon	_____	f. Is there any other element with a larger atomic number?

Chapter 5 Review The Periodic Law Mixed Answer Key

Daniela Niemeyer



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