

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfide, the roman numeral tells us (2)
 - How many atoms of Fe are in one formula unit.
 - How many sulfide ions can be attached to the iron atom.
 - The charge on each Fe ion.
 - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is (1)
 - Change the number of ions represented in the formula.
 - Change the charges on the other ions in the compound.
 - Change the formula so that it no longer represents the compound.
 - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

| Compound Name | Formula |
|--------------------------|----------------------------------|
| Acetic acid | $\text{CH}_3\text{CO}_2\text{H}$ |
| Aluminum sulfate | $\text{Al}_2(\text{SO}_4)_3$ |
| Lead II carbonate | PbCO_3 |
| Ammonium phosphate | $(\text{NH}_4)_3\text{PO}_4$ |
| Sulfuric acid | H_2SO_4 |
| Nitrogen pentoxide | N_2O_5 |
| Copper II carbonate | CuCO_3 |
| Sodium oxide | Na_2O |
| Ammonium phosphate | $(\text{NH}_4)_3\text{PO}_4$ |
| Mercury sulfate | HgSO_4 |
| Hydroic acid | HNO_3 |
| Magnesium perchlorate | $\text{Mg}(\text{ClO}_4)_2$ |
| Iron II nitrate | $\text{Fe}(\text{NO}_3)_2$ |
| Iron III nitrate | $\text{Fe}(\text{NO}_3)_3$ |
| Carbon Dioxide | CO_2 |
| Sulfur dioxide | SO_2 |
| Sulfur pentoxide | SO_5 |
| Tetraphosphorus hexoxide | P_4O_6 |
| Tetraphosphorus decoxide | P_4O_{10} |

- Name the following compound and identify its molecular ions or acid.
 - $\text{Ca}(\text{NO}_3)_2$, calcium (II) nitrate - II
 - $\text{Fe}(\text{NO}_3)_3$, Iron (III) nitrate - III
 - $\text{Fe}(\text{OH})_2$, Iron (II) hydroxide - II
 - N_2O_5 , Nitrogen pentoxide - III

Chapter 9 Chemical Names And Formulas Section Review Answers

Fred M. Dewey



Chapter 9 Chemical Names And Formulas Section Review Answers:

Understanding Chemistry Fred M. Dewey, 1994 **Essential Concepts of Chemistry** Sharon Sherman, Alan Sherman, 1999 Designed especially for students who have little or no background in chemistry or mathematics *Essential Concepts of Chemistry* makes complex concepts understandable This text provides an inexpensive one color alternative for introductory chemistry courses and emphasizes everyday applications of chemistry **Chemistry** Henry Dorin, 1987
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