

## Thermodynamic Equilibrium - 1 (Application of Equilibrium)

### Thermodynamic Equilibrium

Thermodynamic Equilibrium: A system is in thermodynamic equilibrium if its macroscopic properties do not change with time.

1. In a closed system at constant volume and pressure, which of the following is not a state function?
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
2. For the thermodynamic system shown below, which process is not a state function?
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
3. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Entropy
  - C. Heat
  - D. None of the above
4. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Entropy
  - C. Heat
  - D. None of the above
5. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Entropy
  - C. Heat
  - D. None of the above
6. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
7. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
8. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
9. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above
10. Which of the following is not a state function? (Enthalpy, Entropy, Heat, Work)
- A. Enthalpy
  - B. Heat
  - C. Entropy
  - D. All of the above

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**G Psacharopoulos**



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