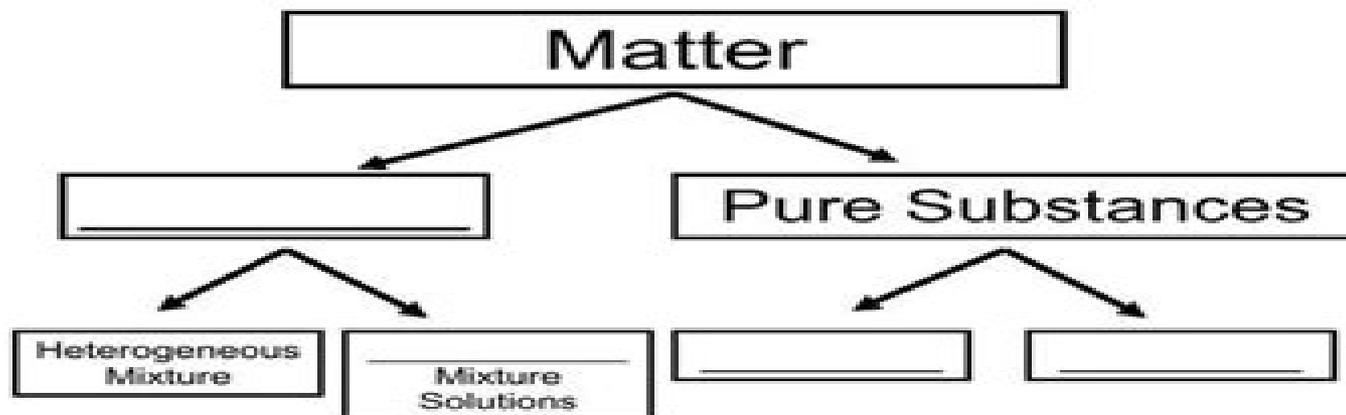


## Chapter 2 Section 1



### Matter

Anything that \_\_\_\_\_

All the "stuff" around you is matter

Matter can be divided into two categories: \_\_\_\_\_ and \_\_\_\_\_

### Pure Substances

\_\_\_\_\_ that are a single kind of matter

Always has the same make up and \_\_\_\_\_ no matter where it comes from

Ex: table salt- always comes from seawater or a salt mine

### Elements

A pure substance that \_\_\_\_\_ be broken down into any other substance

They are the \_\_\_\_\_ substance

Ex: carbon, calcium

### Compound

Made up of \_\_\_\_\_ elements chemically combined

Ex: Water- hydrogen and oxygen

### Mixtures

Made up of two or more substances ( \_\_\_\_\_ or \_\_\_\_\_ but they

DO NOT chemically combine

They \_\_\_\_\_ be separated

### Heterogeneous Mixture

You can see the \_\_\_\_\_ parts

\_\_\_\_\_ the parts easily

Ex: Salad or sand and flour

### Homogenous Mixture - Solutions

A very well \_\_\_\_\_ mixture

Can't see or tell that there are \_\_\_\_\_ parts even though there are

Ex: sugar water – can't tell the sugar from the water after mixing

# Chemistry Matter And Change Chapter 13 Answer Key

**James E. Brady, John R. Holum**



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