

MULTIPLE CHOICE QUESTIONS (MCQ)

1. Which of the following is dependent of temperature ?
(a) Molarity (b) Molality
(c) Mole fraction (d) Mass percentage
2. 4 g of NaOH dissolved in 100 ml solution. Molarity of the solution is
(a) 1 M (b) 10 M
(c) 0.1 M (d) 4 M
3. Which has the maximum number of molecules among the following ?
(a) 44g of CO_2 (b) 44g of O_2
(c) 8g of H_2 (d) 64g of SO_2
4. 10 mol of Zn react with 10 mol of HCl. Calculate the number of moles of H_2 produced.
(a) 5 mol (b) 10 mol
(c) 20 mol (d) 2.5 mol
5. The number of oxygen atoms in 4.4g of CO_2 is approximately
(a) 1.2×10^{23} (b) 6×10^{22}
(c) 6×10^{23} (d) 12×10^{23}
6. The molarity of a solution obtained by mixing 750 mL of 0.5 M HCl with 250 ml of 2 M HCl will be
(a) 0.975 M (b) 0.875 M
(c) 1.00 M (d) 1.175 M
7. Number of atoms of He in 100 u of He (Atomic mass of He is 4 u)
(a) 25 (b) 50
(c) 100 (d) 400
8. 6.02×10^{20} molecules of urea are present in 100 mL of its solution. The concentration of the solution is
(a) 0.02 M (b) 0.01 M
(c) 0.001 M (d) 0.1 M

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Thomas Griffiths



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