

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
2. What colour do sodium ions go in a flame test?
 - a) Red
 - b) Yellow
 - c) Brick-red
 - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and hydrochloric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Damp blue litmus paper goes red
 - c) Damp red litmus paper goes blue
 - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
6. Which ion produces a brick-red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
7. What is the test for carbon dioxide gas?
 - a) Limewater turns cloudy
 - b) Damp blue litmus turns red
 - c) A glowing splint relights
 - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - a) White
 - b) Blue
 - c) Green
 - d) Red/brown
9. What is the test for chlorine gas?
 - a) Damp blue litmus paper turns red and is then bleached white
 - b) Red litmus turns blue
 - c) A glowing splint relights
 - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
 - a) Blue
 - b) Green
 - c) Red/brown
 - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and sulphuric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
 - a) Lilac
 - b) Red
 - c) Yellow
 - d) Brick-red
13. What is the test for hydrogen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Glowing splint relights
 - c) Damp blue litmus paper turns red and then is bleached white
 - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
15. What colour precipitate do bromide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
16. What needs to be added to test for the presence of carbonate ions?
 - a) Silver nitrate
 - b) Universal indicator
 - c) Litmus paper
 - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - a) Oxygen
 - b) Ammonia
 - c) Carbon dioxide
 - d) Hydrogen
18. What colour precipitate does barium sulphate form?
 - a) White
 - b) Green
 - c) Yellow
 - d) Cream
19. What is needed to carry out a flame test?
 - a) Splint
 - b) Glass rod
 - c) Spatula
 - d) Nichrome/platinum wire

Your score out of 19=
Percentage =

Chemistry Unit 2 Test Answers

M Tight



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