

E1	E2	E3
Volume = 1 unit	Volume = 2 units	Volume = 3 units
External pressure = 1 atm	External pressure = 0.50 atm	External pressure = 0.33 atm
Internal pressure = 1 atm	Internal pressure = 0.50 atm	Internal pressure = 0.33 atm
Temperature = 200 K	Temperature = 200 K	Temperature = 200 K

12. Consider the gas samples in Model 2.

a. Name two materials that the containers in Model 2 could be made from that would ensure that they were "flexible"?

The container could be made of Rubber or Latex.

b. What is always true for the external and internal pressures of a gas in a flexible container?

The external and internal pressures are equal in a flexible container



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**Ohio Gas Light Association**



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