

CHEMFILE MINI-GUIDE TO PROBLEM SOLVING**General Plan for Solving Dilution Problems**

$$1 \quad M_1V_1 = M_2V_2$$

Rearrange the equation $M_1V_1 = M_2V_2$ algebraically to solve for the unknown quantity.

2

The equation used to calculate the unknown quantity will be one of the following four:

$$V_2 = \frac{M_1V_1}{M_2}, \quad M_2 = \frac{M_1V_1}{V_2}, \quad V_1 = \frac{M_2V_2}{M_1}, \quad M_1 = \frac{M_2V_2}{V_1}$$

Substitute each of the known quantities for its symbol, and calculate.

$$3 \quad \text{Unknown molarity or volume}$$

SAMPLE PROBLEM 1

What is the molarity of a solution that is made by diluting 50.00 mL of a 4.74 M solution of HCl to 250.00 mL?

SOLUTION**1. ANALYZE**

- *What is given in the problem?* the molarity of the stock solution, the volume used to dilute, and the volume of the diluted solution
- *What are you asked to find?* the molarity of the diluted solution

Items	Data
Concentration of the stock solution (M_1)	4.74 M HCl
Volume of stock solution used (V_1)	50.00 mL
Volume of diluted solution (V_2)	250.00 mL
Concentration of the diluted solution (M_2)	? M

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M Planty



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Holt Chemistry File, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

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Steve Russo, Mike Silver Sandra Yancy McGuire, Steve Russo, Mike Silver, 2002 Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills

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