

Chemistry Worksheet Class 11 on Chapter 13 Hydrocarbons - Set 1

Q-1: Which of the following is the chemical formula of marsh gas?

- a) C_2H_6
- b) CH_4
- c) C_2H_2
- d) C_4H_{10}

Q-2: What would the alkane's formula be if one of the hydrogen atoms in propane was replaced by an ethyl group?

Q-3: Calculate the number of sigma and pi bonds in the following compounds:

- a) 4-Ethyl-2-methylhexane
- b) 3,3-Diethyl-5-isopropyl-4-methyloctane
- c) 4-Ethyl-2,5,7-decatriene

Q-4: Give the IUPAC name for the following:

- a) $CH_3-CH=CH_2$
- b) $CH_2=C(CH_3)_2$
- c) $CH_3-CH_2-C(CH_3)=CH_2$
- d) $CH_2=CH-CH=CH_2$
- e) $CH_3-CH_2-CH(CH_2-CH_3)-CH_2-CH(CH_3)-CH_2-CH_3$

Q-5: What happens to the electron cloud of carbon-hydrogen bonds in an ethane molecule if the configuration is changed from staggered to eclipsed?

Q-6: Which of the following conformations has hydrogen atoms attached to two carbon atoms as close together as possible?

- a) Staggered
- b) Eclipsed
- c) Skew
- d) Gauche

Q-7: What type of isomerism do the compounds listed below exhibit?

Chapter 13 Chemistry Test Answers

Jack Crabtree



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