

Chapter 9 Practice Problems: Chemical Names and Formulas

Section 9.1: Naming Ions

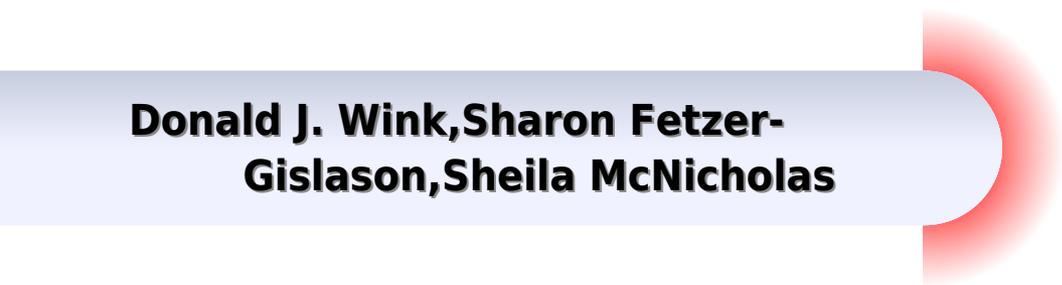
- What is the charge on the ion typically formed by each element?
 - oxygen
 - iodine
 - sodium
 - aluminum
 - nickel, two electrons lost
 - magnesium
- How many electrons does the neutral atom gain or lose when each ion forms?
 - Cr^{3+}
 - P^{3-}
 - Li^+
 - Ca^{2+}
 - Cl^-
 - O^{2-}
- Name each ion. Identify each as a cation or an anion.
 - Sn^{2+}
 - Co^{3+}
 - Br^-
 - K^+
 - H^-
 - Mn^{2+}
- Write the formula (including the charge) for each ion. Use table 9.3 as if necessary.
 - carbonate ion
 - nitrite ion
 - sulfate ion
 - hydroxide ion
 - chromate ion
 - ammonium ion
- Name each of the following ions. Identify each as a cation or an anion.
 - CN^-
 - HCO_3^-
 - PO_4^{3-}
 - Cl^-
 - Ca^{2+}
 - SO_3^{2-}

9.2 Naming and Writing Formulas for Ionic Compounds:

- Write the formulas for these binary ionic compounds.
 - magnesium oxide
 - tin(II) fluoride
 - potassium iodide
 - aluminum chloride
 - sodium sulfide
 - ferric bromide
- Write the formulas for the compounds formed from these pairs of ions.
 - Ba^{2+} and Cl^-
 - Ag^+ and I^-
 - Ca^{2+} and S^{2-}
 - K^+ and Br^-
 - Al^{3+} and O^{2-}
 - Fe^{2+} and O^{2-}
- Name the following binary ionic compounds.
 - MnO_2
 - Li_3N
 - CaCl_2
 - SrBr_2
 - NiCl_2
 - K_2S
 - Fe_3N_2
 - SnCl_4
- Write formulas for the following ionic compounds.
 - sodium phosphate
 - magnesium sulfate
 - sodium hydroxide
 - potassium cyanide
 - ammonium chloride
 - potassium dichromate
- Write formulas for the following ionic compounds.
 - NH_4^+ and SO_4^{2-}
 - K^+ and NO_3^-
 - barium ion and hydroxide ion
 - lithium ion and carbonate ion

Chapter 9 Chemical Names Formulas Practice Problems Answers

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



Chapter 9 Chemical Names Formulas Practice Problems Answers:

Chapter 9 Chemical Names Formulas Practice Problems Answers Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Chapter 9 Chemical Names Formulas Practice Problems Answers**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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kain perca juga beragam dari kain perca kecil hingga besar sedangkan kerajinan dari kain perca merupakan sebuah hasil kreasi yang dihasilkan dari sisa kain perca tersebut

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web kain perca adalah istilah untuk limbah atau sisa sisa kain jenis kain ini dapat dibuat menjadi kerajinan dari limbah kain perca sebageian orang menilai bahwa kain perca adalah sampah yang tidak bisa digunakan lagi karena kain ini hanyalah sisa sisa dari pembuatan pakaian

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