

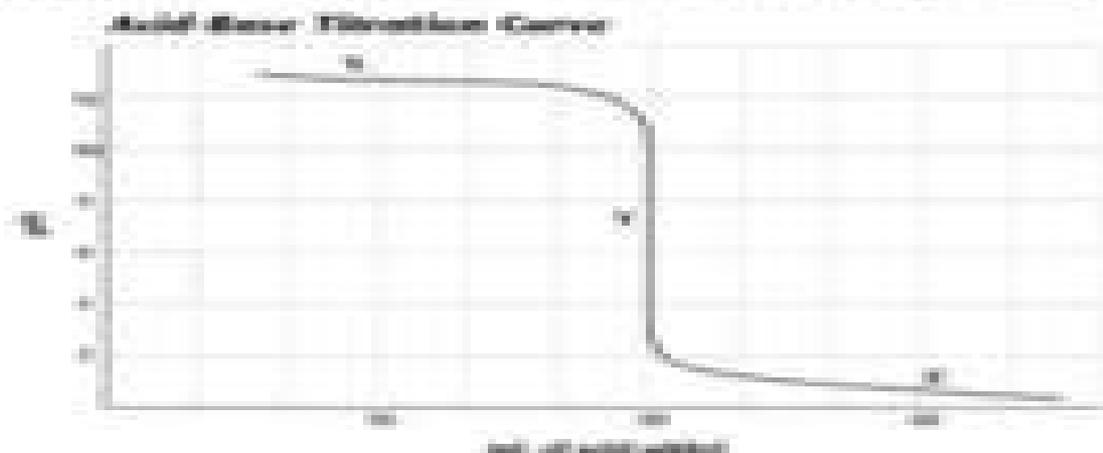
CHAPTER 14 REVIEW

Acid-Base Titration and pH

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Below is a pH curve from an acid-base titration. The x-axis labels show points A, B, C, and D.



- _____ a. Which point represents the equivalence point?
- _____ b. At which point is there excess acid in the system?
- _____ c. At which point is there excess base in the system?
- _____ d. If the three solutions in (b)-(c) are acid-base neutralizations of 100 mL each, how much acid or base, from every reaction of 100 mL, does not react at the end point of the reaction?

PROBLEM Write the answer on the line to the left. Show all your work in the space provided.

1. A 100-mL solution of 0.100 M HCl is mixed with a 100-mL solution of sodium hydroxide to decrease its acidity and its volume. It takes 25 mL of the base to neutralize the 100 mL of the acid.
- a. Write the balanced molecular equation for this neutralization reaction.

Chapter 15 Acid Base Titration Ph Section 2 Answers

Christie G. Enke



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