

STOICH.1	Balance chemical equations by applying the laws of conservation of mass and constant composition.	9.1
STOICH.3	Use mole ratios to describe the relationship between coefficients and moles in a chemical reaction.	11.1
STOICH.4	Use dimensional analysis to calculate amount of reactants necessary and products produced in terms of moles, grams and particles.	11.2

Use Stoichiometry to solve the following problems. Use the conversion chart to help.



A is starting material and B is ending material

1. Magnesium reacts with hydrochloric acid according to the following chemical equation:



If two moles of hydrochloric acid react with excess magnesium, how many moles of hydrogen gas will be produced?

- What is your starting point? _____
- What are the units on your starting point? _____
- What is your ending point? _____
- What are the units on your ending point? _____
- Balance the equation.
- Solve the problem:

2. Aluminium reacts with HCl to produce aluminum chloride and hydrogen gas.

- Write a balanced equation for the reaction.
- Calculate the number of moles of HCl required to react with 0.87 moles of Al.

3. Glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) combines with oxygen in the body in a combustion reaction to form carbon dioxide and water. Balance the reaction.



How many moles of oxygen are needed to react with the 29 grams of glucose in a Milky Way candy bar?

- What is your starting point? _____
- What are the units on your starting point? _____
- What is your ending point? _____
- What are the units on your ending point? _____
- Solve the problem:

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Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor's from Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master's University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

The Art of Teaching Science Jack Hassard, Michael Dias, 2013-07-04 The Art of Teaching Science emphasizes a humanistic experiential and constructivist approach to teaching and learning and integrates a wide variety of pedagogical tools Becoming a science teacher is a creative process and this innovative textbook encourages students to construct ideas about science teaching through their interactions with peers mentors and instructors and through hands on minds on activities designed to foster a collaborative thoughtful learning environment This second edition retains key features such as inquiry based activities and case studies throughout while simultaneously adding new material on the impact of standardized testing on inquiry based science and explicit links to science teaching standards Also included are expanded resources like a comprehensive website a streamlined format and updated content making the experiential tools in the book even more useful

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