



The Chemistry of Life

Chapter Test A

Multiple Choice

Directions: Select the best answer for each question by circling the number of the choice provided.

- _____ 1. If an atom has a mass number of 23 and 10 electrons, the atom has a net charge of _____.
- a. 13
b. 10
c. 11
d. 12
- _____ 2. If an atom has a mass number of 23 and 10 protons, the atom has a net charge of _____.
- a. 13
b. 10
c. 11
d. 12
- _____ 3. Which of the following would be considered an acid?
a. low amount of hydrogen and low amount of oxygen
b. low amount of hydrogen and low amount of oxygen
c. low amount of hydrogen and low amount of oxygen
d. low amount of hydrogen and low amount of oxygen
- _____ 4. A molecule that is neutral is the molecule
a. containing electrons
b. containing protons
c. having electrons
d. having protons
- _____ 5. Which molecule can polarize water?
a. oxygen with being slightly positive and the hydrogen with being slightly negative
b. oxygen with hydrogen with being slightly positive
c. oxygen with hydrogen with being slightly negative
d. oxygen with being slightly negative and the hydrogen with being slightly positive
- _____ 6. If you are with water, which is the most polar?
a. positive with negative
b. positive with negative
c. positive with negative
d. positive with
- _____ 7. A substance with a pH of 10 is called
a. an acid
b. a base
c. a salt
d. a compound
- _____ 8. Sodium chloride has a pH of 7. It is called
a. an acid
b. a base
c. a salt
d. a mixture
- _____ 9. Why is coffee considered compared to other drinks?
a. Coffee is acidic and has a low pH and it has a lot of different nutrients.
b. Coffee is acidic and has a low pH and it has a lot of different nutrients.
c. Coffee is acidic and has a low pH and it has a lot of different nutrients.
d. Coffee is acidic and has a low pH and it has a lot of different nutrients.

Chapter Test Chemistry Of Life Answer Key

Gary A. Thibodeau, Kevin T. Patton



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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993 **Chemistry and Life** John W. Hill, 1997

Life William K. Purves, 2001 Authoritative thorough and engaging Life The Science of Biology achieves an optimal

balance of scholarship and teachability never losing sight of either the science or the student The first introductory text to present biological concepts through the research that revealed them Life covers the full range of topics with an integrated experimental focus that flows naturally from the narrative This approach helps to bring the drama of classic and cutting edge research to the classroom but always in the context of reinforcing core ideas and the innovative scientific thinking behind them Students will experience biology not just as a litany of facts or a highlight reel of experiments but as a rich coherent discipline

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Psychology in Everyday Life (High School) David G. Myers,C. Nathan DeWall,2017-01-27 The psychology course is often a student favorite because it s so relevant to our everyday lives And like all Myers and DeWall texts *Psychology in Everyday Life* delivers an enjoyable and entertaining read complete with the latest research and an eye catching design Unlike other Myers DeWall books this is a brief offering that makes no assumptions about a students background or experiences This book is written with the diversity of your students in mind **Glencoe Science** ,2002 **Inquiry Into Life** Leslie J. Wiemerslage,Sylvia S. Mader,1994-02 Basic biological concepts and processes with a human emphasis From the unique delivery of biology content to the time tested art program to the complete integration of the text with technology Dr Sylvia Mader has formed a teaching system that will both motivate and enable your students to understand and appreciate the wonders of all areas of biology *Inquiry into Life 12 e* emphasizes the application of all areas of biology to knowledge of human concerns what the students are able to relate to This distinctive text was developed to stand apart from all other non majors texts with a unique approach unparalleled art and a straightforward succinct writing style that has been acclaimed by both users and reviewers *Study Guide for Life on Earth* Carl S. Pike,1973 **Structure & Function of the Body** Gary A. Thibodeau,Kevin T. Patton,2000 *Structure and Function of the Body 11 The 11 Books in Print Supplement* ,2002

Human Biology Sylvia S. Mader,1997-07 **Biochemistry** Peck Ritter,1996 *Psychology in Everyday Life* David G. Myers,2008-10-17 LEARN IT LIVE IT Why take psychology What makes psychology a science Can it really help me understand my feelings and behaviors Or how I get along with family and friends Now from the world s foremost author for the introductory psychology classroom comes a new textbook that makes learning about the psychology of our lives a captivating experience for students at all levels Carried by the author s acclaimed empathetic voice *Psychology in Everyday*

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