

CHEMISTRY TEST

NAME: _____

INTRODUCTORY CHEMISTRY CONCEPTS

DATE: _____

Directions: For each of the following questions, choose the number that **best** answers the question and place it on your answer sheet.

Directions: For each of the following questions, all major steps must be shown to receive full credit.

Introductory Chemistry Concepts**Level 1****Multiple Choice**

1. A chemical change always results characteristically in a change of

- 1) state; 2) form; 3) identity; 4) size; 5) shape.

ANS: 3

2. Which of the following is a chemical change?

- 1) evaporation of water 2) distillation of water 3) freezing of water
4) electrolysis of water

ANS: 4

3. Chemical changes always involve 1) the absorption of energy; 2) the liberation of energy;
3) either the absorption or liberation of energy; 4) changes of state.

ANS: 3

4. One physical property of sugar is its 1) acidity; 2) reactivity;

- 3) flammability; 4) plasma qualities; 5) solubility.

ANS: 5

5. The majority of chemical changes are 1) autothermic; 2) endothermic;

- 3) isothermic; 4) nonthermic; 5) exothermic.

ANS: 5

6. The statement which describes a **physical** property of the element oxygen is

- 1) oxygen supports the burning of paper; 2) oxygen is needed for the rusting of iron;
3) oxygen must be present for food to decay; 4) oxygen has a density of 1.43 grams per liter;
5) oxygen is a component of water.

ANS: 4

7. In any chemical reaction energy is 1) always absorbed; 2) always

- released; 3) released as heat; 4) usually not involved; 5) either absorbed or released.

ANS: 5

8. Which of the following is **not** a chemical change?

- 1) the dissolving of zinc by HCl 2) the exploding of gasoline 3) the
corroding of a copper roof 4) the boiling of water

ANS: 4

9. One chemical property of a substance is a description of its 1) odor; 2) density;

- 3) combustibility; 4) solubility; 5) none of these.

ANS: 3

10. A new substance is formed as a result of a 1) physical change; 2) chemical

Chemistry Matter Change Section Assessment Answers

Leo J. Malone, Theodore O. Dolter



Chemistry Matter Change Section Assessment Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28
The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon, buckyballs, graphite, and diamond, are illustrated at the left, as is the molecule methane (CH₄) from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement.

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21
This Third Edition revised to provide smoother transitions between topics, employs a concise yet informal approach to basic chemistry, organized to help students employ basic math skills and problem-solving strategies. Writing style is straightforward, and presentation incorporates many concrete analogies to clarify new concepts. Includes many illustrative worked examples.

Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08
Glencoe Chemistry Solving Problems: A Chemistry Handbook: Matter and Change

The Veterinary Technician United States. Department of the Air Force, 1968

AF Manual United States. Department of the Air Force, 1968

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004
The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even-numbered end-of-chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.

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accompany Basic Concepts of Chemistry, 7th Edition Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry *School Science and Mathematics* ,1919 **Chemical Engineering Catalog** ,1921

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