

Chapter Test A

Chemical Bonding

MULTIPLE CHOICE

Write the letter of the correct answer in the space provided.

- _____ 1. How are electrons in an atom organized?
- energy levels
 - valence levels
 - groups
 - periods
- _____ 2. What are the only parts of an atom that make chemical bonds?
- protons in the nucleus
 - neutrons in the nucleus
 - electrons in the nucleus
 - valence electrons
- _____ 3. What is the same within each group on the periodic table?
- number of protons
 - number of electrons
 - number of atoms
 - number of valence electrons
- _____ 4. Why don't noble gases normally form chemical bonds?
- They have 2 valence electrons.
 - They have full outermost energy levels.
 - They have empty outer levels.
 - They have 8 protons.
- _____ 5. What is the charge of a sodium ion with 11 protons and 10 electrons?
- 1+
 - 2+
 - 1-
 - 2-
- _____ 6. The more easily an atom gains an electron, what does it release more of?
- atoms
 - energy
 - electrons
 - protons

Chapter 13 Practice Test

AW Chickering



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