

## Chemistry Student Edition - Basic Answer Key

### Chapter 11: Chemical Reactions

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#### Chemical Equations

##### Questions

1. Early Chinese alchemists discovered an early form of gunpowder. What was the composition of this substance?
2. What later developments were made to the gunpowder recipe that improved its pyrotechnic properties?
3. Make an argument for why the burning of a candle is consistent with the law of conservation of matter/mass.
4. Think of an experiment that you could conduct to demonstrate that mass is conserved for a given chemical change.
5. Balance the following chemical equations:
  - a.  $C + O_2 \rightarrow CO$
  - b.  $CO + O_2 \rightarrow CO_2$
  - c.  $H_2 + Br_2 \rightarrow HBr$
  - d.  $K + H_2O \rightarrow KOH + H_2$
  - e.  $O_3 \rightarrow O_2$
  - f.  $N_2 + H_2 \rightarrow NH_3$
  - g.  $Zn + AgCl \rightarrow ZnCl_2 + Ag$
  - h.  $Cl_2 + NaI \rightarrow NaCl + I_2$
  - i.  $P_4O_{10} + H_2O \rightarrow H_3PO_4$
  - j.  $Be_2C + H_2O \rightarrow Be(OH)_2 + CH_4$
  - k.  $S + HNO_3 \rightarrow H_2SO_4 + NO_2 + H_2O$
  - l.  $NH_3 + CuO \rightarrow Cu + N_2 + H_2O$
  - m.  $HCl + CaCO_3 \rightarrow CaCl_2 + H_2O + CO_2$

##### Answers

1. Early gunpowder was comprised of saltpeter (impure potassium nitrate), sulfur, and charcoal (carbon).
2.
  - a. Purification of potassium nitrate
  - b. Using potassium chlorate instead of potassium nitrate
  - c. Further refinement of the recipe to find the proportions of reactants that yielded the most energetic reaction
3. The burning of a candle obeys the law of conservation of mass because, even though the mass of the candle will decrease as it burns, the mass is not lost. It is transferred into carbon dioxide and water that enter the surrounding air.
4. The reaction between iron and oxygen to produce rust could be studied as an example of conservation of mass. Steel wool could be placed into a covered beaker containing water. The reaction would proceed over time and rust would begin to form. The beaker could be weighed before and after the change. The mass would be the same, even though a new

# Chapter 11 Chemical Reactions Answer Key

**Terry C. Jones**



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In some sort of used by monitors and the ceaseless chatter of instantaneous transmission, the melodic elegance and mental symphony created by the written term frequently disappear into the background, eclipsed by the constant noise and disturbances that permeate our lives. However, set within the pages of **Chapter 11 Chemical Reactions Answer Key** a marvelous fictional treasure overflowing with organic emotions, lies an immersive symphony waiting to be embraced. Crafted by a wonderful composer of language, that charming masterpiece conducts visitors on an emotional trip, skillfully unraveling the hidden songs and profound impact resonating within each carefully constructed phrase. Within the depths of the emotional examination, we shall examine the book is key harmonies, analyze their enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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