

Chapter 6 Chemical Bonds

WordWise

Unscramble the terms from the following list to fit each of the clues given below.

- | | | |
|------------|----------|----------|
| daimlet | ecumelol | levoctan |
| lorpa | loyal | marfulo |
| mooctyliap | nocii | nonia |
| odbn | starscly | tonica |

Clues

- A type of bond that holds cations and anions together
- A type of bond in which two atoms share a pair of valence electrons
- A neutral group of atoms that are joined together by one or more covalent bonds
- A term describing a covalent bond in which electrons are not shared equally
- An ion that contains a covalently bonded group of atoms
- An ion with a negative charge
- An ion with a positive charge
- A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound
- Solids whose particles are arranged in a lattice structure
- A mixture of two or more elements, at least one of which is a metal
- A type of bond that exists between a metal cation and the shared electrons that surround it
- The force that holds atoms or ions together

Vocabulary Terms

Chapter 6 Chemical Bonds Wordwise

Sabine Zange

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

Chapter 6 Chemical Bonds Wordwise:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11
Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Chemical Bonds* Phillip Manning,2009 Modern life is made up of a mind boggling array of materials A simple drinking cup for example might be made of Styrofoam paper or glass depending on the drinkers needs at the moment Home storage cabinets can be made of metal wood or plastic Space shuttles are assembled from silicon steel and hundreds of other materials All of these items owe their properties to the chemical bonds between the atoms that make up the substance *Chemical Bonds* examines the nature of the chemical bonds answering fundamental questions about how they form how they are broken and how they help define life as we know it *Chemical Bonding Clarified Through Quantum Mechanics* George C. Pimentel,Richard D. Spratley,1969 Chemical Bonds and Bonds Energy R Sanderson,2012-12-02 *Chemical Bonds and Bonds Energy* Second Edition provides information pertinent to the fundamental aspects of contributing bond energy and bond dissociation energy This book explores the values that are useful in the interpretation of significant phenomena such as product distribution and reaction mechanisms Organized into 12 chapters this edition begins with an overview of the quantitative relationship among three basic properties of an atom namely nonpolar covalent radius electronegativity and homonuclear single covalent bond energy This text then examines the quantitative means of evaluating the partial atomic charges that result from initial differences in the electromagnetivity of atoms that form a compound Other chapters consider the recognition of the reduction of bond weakening not by multiplicity and in certain types of single covalent bonds The final chapter deals with the application of the principal ideas and techniques to the oxidation of ethane This book is a valuable resource for organic and inorganic chemists *Chemical Bonds* Jeremy K. Burdett,1997-05-21 Taking a question and answer format this tutorial explores the fundamental concepts of chemical bonding and is based on the lectures of Professor Jeremy Burdett Chemical Bonds ,2010 **Chemical Bonding** Audrey L. Companion,1979 Chemical Bonds and Bond Energy Robert Thomas Sanderson,1971 The Nature of the Chemical Bond and the Structure of Molecules and Crystals Linus Pauling,1963

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