

# Chemistry Applications Study Guide

## Define chemical reaction

A Chemical Reaction is a process where two or more substances interact with each other, causing their atoms to rearrange and form new substances with different properties

## Define Physical Change:

Physical Change is when matter changes forms but not chemical identity

## Give examples of physical changes

- Change of State ( Solid, Liquid, Gas)
- Dissolving ( Sugar in Water )
- Changing the Size or Shape of an Object ( Crushing a Can )

## Define Chemical Change:

During a Chemical Change a Chemical Reaction Occurs, New Substances Form, Chemical Reactions Produce new Substances by Changing the way Atoms are Arranged

## Give examples of chemical changes:

- Rusting ( The Statue of Liberty )
- Burning ( A Campfire )

## Define reactants and products:

Reactants:  
What you start with, bonds are broken  
Products:  
What you end with, atoms are rearranged

## List six types of evidence for Chemical Reactions

- Color Change
- Light Emission
- Formation of a Gas

# Chemistry Concepts Applications Study Guide Answer

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