

## Stoichiometry Worksheet 2

1. Sodium chloride can be produced by the reaction of sodium and chlorine gas. Write a balanced equation for this reaction.

Suppose that 6.70 mol Na react with 3.20 mol  $\text{Cl}_2$ .

- What is the limiting reagent?
- How many moles of NaCl are produced?

2. The equation for the complete combustion of ethane,  $\text{C}_2\text{H}_6$ , is



Balance the equation. If 2.70 mol  $\text{C}_2\text{H}_6$  is reacted with 6.30 mol  $\text{O}_2$ ,

- identify the limiting reagent.
  - calculate the moles of water produced.
3. Copper and sulfur combine to form copper (I) sulfide.
- $$2 \text{Cu} (\text{s}) + \text{S} (\text{s}) \rightarrow \text{Cu}_2\text{S} (\text{s})$$
- What is the limiting reagent when 80.0 g Cu reacts with 25.0 g S?
  - What is the maximum number of grams of  $\text{Cu}_2\text{S}$  that can be formed?

# Chemistry Stoichiometry Problem Sheet 2 Answer Key

**ML Morrison**



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