

## BSC 1005 Exam 2 review questions with answers

1. **First Law of Thermodynamics:** energy cannot be created or destroyed energy is converted or transformed from one form to another
2. **Second Law of Thermodynamics:** energy transformation is not 100%. In energy transformation some energy is lost as heat. Heat is a random and unordered form of energy that cells cannot do too much with.
3. **Entropy:** the tendency of a system toward randomness or disorganization

the universe's entropy is increasing

4. **Potential Energy:** stored energy or energy that has to do with the position that is available to do work.

ex- chemical energy, ATP

5. **Kinetic Energy:** energy associated with motion or movement; is the energy being used to do work.

Heat is related to kinetic energy. It is a very random form of energy and cells cannot do too much with heat

6. **Oxidation Reaction:** losing electrons
7. **Reduction Reaction:** gaining electrons
8. **Exergonic Reaction:** chemical reaction that proceeds with a net release of energy. This type of chemical reaction is normally used to break down complex molecules. Cellular respiration is an example of an exergonic process
9. **Endergonic Reaction:** reaction that requires an input of energy in order to convert reactants (substrates) into products. This type of chemical reaction is often used to build complex molecules. Photosynthesis is an example of an endergonic process.
10. **what is ATP:** adenosine triphosphate is the universal energy carrier in cells and in all living organisms. It is an organic molecule; in fact it is a modified nucleotide.

has an adenine nitrogenous base, a ribose sugar and three phosphate groups instead of only one.

11. **how does ATP work:** The way ATP provides energy to different events is by losing phosphate groups with the release energy. This process is exergonic (i.e. it releases energy; see below for the hydrolysis of ATP). The synthesis of ATP (i.e the reverse process) is endergonic. Addition of a phosphate group to ADP forms ATP.

# Bsc 1005 Laboratory Manual Answers Yuexiangore

**Raffaela Di Napoli**



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