

## Aqueous Reactions: Worksheet #1

### Answers

1.

- a. No reaction  
 b. Reaction ( $\text{PbCl}_2$  - ppt)  
 c. Reaction (HOH - liquid)  
 d. Reaction ( $\text{CaSO}_4$  - ppt)

2.

*All spectator**All spectator**All spectator*

3. Multiple methods. The basic idea is that 2 soluble chemicals are used, one with a silver ion ( $\text{Ag}^+$ ) and the other with a chromate ion ( $\text{CrO}_4^{2-}$ ). Be sure however, that the other product remains soluble.



4.



5.

- a. Hydrochloric Acid  
 b. Hydrobromic Acid  
 c. Hydroiodic Acid  
 d. Sulphuric Acid  
 e. Nitric Acid  
 f. Perchloric Acid

# Reactions In Aqueous Solution Worksheet Answers

**IM Harris**



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