

Stoichiometry

Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- _____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- _____ 2. Stoichiometry is based on the law of conservation of mass.
- _____ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- _____ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- _____ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen \rightarrow carbon dioxide + water



Substance	Molar Mass (g/mol)	Number of Molecules	Number of Moles (mol)	Mass (g)
6. Methanol	32.05			
7. Oxygen gas	32.00			
8. Carbon dioxide	44.01			
9. Water	18.02			

10. What are the reactants? _____
11. What are the products? _____
12. What is the total mass of the reactants? _____
13. What is the total mass of the products? _____
14. How do the total masses of the reactants and products compare? _____

Chapter 11 Stoichiometry Study Answer Key

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Chapter 11 Stoichiometry Study Answer Key:

Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Chapter 11 Stoichiometry Study Answer Key**

In a global inundated with screens and the cacophony of immediate interaction, the profound energy and emotional resonance of verbal beauty frequently disappear in to obscurity, eclipsed by the continuous barrage of sound and distractions. Yet, set within the lyrical pages of **Chapter 11 Stoichiometry Study Answer Key**, a fascinating work of literary beauty that pulses with organic thoughts, lies an wonderful journey waiting to be embarked upon. Penned by a virtuoso wordsmith, that exciting opus guides viewers on a mental odyssey, softly revealing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book is central styles, dissect their fascinating writing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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