

## Classes of Chemical Reactions

1. What type of product was formed in A.1.a based on the color of the litmus paper?

Magnesium oxide and water were added to the red litmus paper which caused it to change its appearance to blue, and that indicates its alkaline substances.

2. What is the role of  $\text{FeCl}_3$  in the decomposition of  $\text{H}_2\text{O}_2$ ?

$\text{FeCl}_3$  chloride serves as a catalyst in the decomposition of hydrogen peroxide.

3. The following single replacement reactions are performed with their results below. Determine the relative activities of the two elements involved in each reaction. Then arrange the following elements: gold, hydrogen, zinc, and tin in order of increasing activity.



**Answer:** Gold < Au < Sn < Zn

4. Use your answers in the previous question to write the products for the reactions below. Write "No Reaction" when applicable.



5. Which of the product in Part 4 is the precipitate? Write their formula below:

**Answer:** Part 4:  $\text{Au}(\text{NO}_3)_3$

6. How will you identify if a chemical reaction is a redox reaction?

Redox reaction is said to be a reaction that has change in oxidation number of one of the atoms in the reaction.

# Classes Of Chemical Reactions Lab With Answers

**JS Bruner**



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