

Experiment 11A: The Periodic Table

Procedure

Part 1 Electronic Configuration and the Periodic Table

1. Obtain a sheet of paper.

Part 2 Atomic Number, Atomic Symbol, Electronic Configuration and the Periodic Table

1. Obtain a sheet of paper (graph paper preferred).

Part 3 Electronic Configuration and Energy Levels (Periodic Table)

1. Obtain a sheet of paper.

Background

1. Examine the periodic table of elements and compare it to Figure 11.1. What relationship can be seen between an element's placement in the periodic table and its electronic configuration?

Electronic configurations of the elements are shown in the periodic table of elements and are given with the element. Though data are not shown, each periodic table contains elements:

1. Examine the graph of atomic radii (highlighted) plotted against increasing atomic number (Figure 11.4-4). What is particularly noteworthy about it? Is it identical to the periodic table?

The table provides data on the elements. The data are listed below in graph. Examine the graph and label the horizontal axis as period (row) and the vertical axis as atomic number. The elements are color-coded as follows: noble gases (yellow), alkali metals (red), transition metals (green), lanthanides (blue), actinides (purple), and the remaining elements are the remaining elements.

1. Which groups appear to have elements of the highest atomic radii? Do you group periods? Which group has the smallest?

Examine the graph and label the horizontal axis as atomic number and the vertical axis as period. The noble gases group appears to have the largest atomic radii. Which elements group appears to have the smallest atomic radii?

1. Examine the graph of the first ionization energy plotted against increasing atomic number (Figure 11.4-5). What is particularly noteworthy about it? Is it identical to the periodic table?

It is not quite identical to the periodic table. The noble gases group has the highest ionization energy and the alkali metals group has the lowest ionization energy. The transition metals group has the lowest ionization energy. The noble gases group has the highest ionization energy.

Laboratory Experiment 11a Answers Periodic Table

**Barron's Educational Series, Neil D.
Jespersen, Pamela Kerrigan**



Laboratory Experiment 11a Answers Periodic Table:

Whispering the Secrets of Language: An Mental Journey through **Laboratory Experiment 11a Answers Periodic Table**

In a digitally-driven world wherever screens reign great and quick communication drowns out the subtleties of language, the profound secrets and emotional nuances concealed within phrases frequently go unheard. Yet, located within the pages of **Laboratory Experiment 11a Answers Periodic Table** a fascinating fictional treasure blinking with natural emotions, lies an extraordinary journey waiting to be undertaken. Penned by a skilled wordsmith, that charming opus encourages readers on an introspective journey, gently unraveling the veiled truths and profound affect resonating within ab muscles material of each word. Within the psychological depths of the emotional review, we will embark upon a genuine exploration of the book is primary styles, dissect its charming publishing style, and succumb to the strong resonance it evokes strong within the recesses of readers hearts.

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