

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	Very	Very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (III) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acid Base Theories Section Review Answer

Hussin A.Rothana



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Table of Contents Acid Base Theories Section Review Answer

1. Understanding the eBook Acid Base Theories Section Review Answer
 - The Rise of Digital Reading Acid Base Theories Section Review Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Acid Base Theories Section Review Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acid Base Theories Section Review Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acid Base Theories Section Review Answer
 - Personalized Recommendations
 - Acid Base Theories Section Review Answer User Reviews and Ratings
 - Acid Base Theories Section Review Answer and Bestseller Lists
5. Accessing Acid Base Theories Section Review Answer Free and Paid eBooks
 - Acid Base Theories Section Review Answer Public Domain eBooks
 - Acid Base Theories Section Review Answer eBook Subscription Services
 - Acid Base Theories Section Review Answer Budget-Friendly Options
6. Navigating Acid Base Theories Section Review Answer eBook Formats

- ePub, PDF, MOBI, and More
 - Acid Base Theories Section Review Answer Compatibility with Devices
 - Acid Base Theories Section Review Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acid Base Theories Section Review Answer
 - Highlighting and Note-Taking Acid Base Theories Section Review Answer
 - Interactive Elements Acid Base Theories Section Review Answer
 8. Staying Engaged with Acid Base Theories Section Review Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acid Base Theories Section Review Answer
 9. Balancing eBooks and Physical Books Acid Base Theories Section Review Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acid Base Theories Section Review Answer
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Acid Base Theories Section Review Answer
 - Setting Reading Goals Acid Base Theories Section Review Answer
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Acid Base Theories Section Review Answer
 - Fact-Checking eBook Content of Acid Base Theories Section Review Answer
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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