

# Acid Base Titrations



$$\frac{0.29 \text{ mol HNO}_3}{1} \times \frac{0.045 \text{ L}}{1} \times \frac{1 \text{ mol Ba(OH)}_2}{2 \text{ mol HNO}_3} \times \frac{1 \text{ L}}{0.15 \text{ mol Ba(OH)}_2}$$

$$\times \frac{1000 \text{ mL}}{1} = 43.5 \text{ mL}$$

$$1 M_1 V_1 = 2 M_2 V_2$$

# Acid Base Titration Problems With Answers

**Arnold B. Loebel**



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