

## ACS EXAM (2024 version) Practice Questions and Answers

A 20.0 g sample of mercury(II) oxide ( $\text{HgO}$ ,  $M = 216.6$ )

is heated strongly, causing it to decompose to metallic Hg and  $\text{O}_2$  gas. What volume of  $\text{O}_2$  gas is produced

(measured at STP)?

(A) 1.03 L

(B) 2.07 L

(C) 4.14 L

(D) 14.0 L -  $\checkmark\checkmark$  Correct-A

When 30.0 mL of 0.10 M  $\text{AgNO}_3$  is added to 30.0 mL of

0.10 M  $\text{NaCl}$ , aqueous  $\text{NaNO}_3$  and solid  $\text{AgCl}$  are

formed. How much solid  $\text{AgCl}$  is produced?

(A) 0.0030 mol

(B) 0.0060 mol

(C) 0.030 mol

(D) 0.060 mol -  $\checkmark\checkmark$  Correct-A

How much  $\text{Sr}(\text{OH})_2 \cdot 8 \text{H}_2\text{O}$  ( $M = 265.76$ ) is needed to

# Acs Chemistry Exam Answers

**Joshua Rueda**



**Acs Chemistry Exam Answers:**

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