

## વિભાગ A

- નીચે આપેલા પાઠ જવાબો પૈકી સાચો વિકલ્પ પસંદ કરો. (પ્રત્યેકના 1 મુલ્ય) [50]
- ZnO ના સ્ફટિકને તરમ કરવા તેનું સૂત્ર ..... અને છે.  
(A)  $Zn_{1-x}O$  (B)  $Zn_{1+x}O$  (C) ZnO (D) આપેલ તમામ
  - અંતઃકેન્દ્રિત એકમકોષ રચનાની પેકિંગ ક્ષમતા ..... હોય છે.  
(A) 40% (B) 68% (C) 72% (D) 74%
  - $Fe_xO$  ના  $Fe(II)$  માટે એક  $Fe(III)$  પરમાણુ છે, x નું મૂલ્ય શું હશે ?  
(A)  $\frac{8}{9}$  (B)  $\frac{2}{3}$  (C)  $\frac{3}{4}$  (D)  $\frac{5}{3}$
  - સામાન્ય તાપમાને સોડિયમ પાતુ અંતઃકેન્દ્રિત પદાર્થમાં સ્ફટિકીકરણ પામે છે. એકમ કોષની ધાર  $a = 4.29 \text{ \AA}$  છે. સોડિયમ પરમાણુની સંખ્યા ..... છે.  
(A) 1.40  $\text{ \AA}$  (B) 2.65  $\text{ \AA}$  (C) 1.85  $\text{ \AA}$  (D) 2.15  $\text{ \AA}$
  - પોટાશ એલમનું અણુસૂત્ર .....  
(A)  $K_2SO_4 \cdot Al(SO_4)_3 \cdot 24 H_2O$  (B)  $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24 H_2O$   
(C)  $K_2SO_4 \cdot Al_2(SO_4)_2 \cdot 24 H_2O$  (D)  $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24 H_2O$
  - આબ્જીવ પદાર્થ A અને B ના જલીય દ્રાવણનું સંતુલન સ્થિતિએ કુલ દબાણ 0.02 બાર છે. જો A પદાર્થનો મોલઅંશ 0.2 હોય, તો પદાર્થ B નું અંશિક દબાણ કેટલું હશે ?  
(A) 0.02 બાર (B) 0.04 બાર (C) 0.16 બાર (D) 0.2 બાર
  - 298 K તાપમાને  $CO_2$  ના તેના સંતૃપ્ત જલીય દ્રાવણમાં મોલઅંશ કેટલા હશે ? [ $K_{H_2} = 6.02 \times 10^{-4} \text{ bar}$ ,  $P_{CO_2} = 2 \times 10^{-6} \text{ bar}$ ]  
(A)  $3 \times 10^{-4}$  (B)  $3.32 \times 10^{-5}$  (C)  $1.23 \times 10^{-4}$  (D)  $2.82 \times 10^{-5}$
  - $MgSO_4$ ,  $AgNO_3$  અને  $AlCl_3$  ના દ્રાવણમાંથી 1 મોલ ઈલેક્ટ્રોન પસાર કરતાં કુલો આણત Mg, Ag અને Al નીચેના પૈકી કયા મોલ મુક્તીકરણમાં જમા થશે ?  
(A) 1 : 1 : 1 (B) 1 : 2 : 3 (C) 2 : 1 : 3 (D) 3 : 6 : 2
  - સાંદ  $NaCl$  ના જલીય દ્રાવણનું સેકન્ટન્ટના કુલો વજન વિદ્યુતવિભાજન કરતાં કેથોડ અને એનોડ ઉપર અનુક્રમે શું ઉત્પન્ન થાય છે ?  
(A)  $Cl_2$  અને Na (B)  $H_2$  અને  $O_2$  (C)  $Cl_2$  અને  $H_2$  (D)  $H_2$  અને  $Cl_2$
  - લેડ સંચાલક કોષમાં કયું દ્રાવણ (વિદ્યુતવિભાજન) ભરવામાં આવે છે ?  
(A) 30%(w/w) સાંદ  $H_2SO_4$  (B) 48%(w/w) સાંદ  $H_2SO_4$   
(C) 38%(w/w) સાંદ  $H_2SO_4$  (D) 38%(w/w) સાંદ  $HNO_3$
  - હાર્ડવેજન બળતણ કોષમાં એનોડ પર થતી પ્રક્રિયા કઈ છે ?  
(A)  $2H_{2(g)} + 4OH^{-}_{(aq)} \rightarrow 4H_2O_{(l)} + 4e^{-}$  (B)  $O_{2(g)} + 2H_2O_{(l)} + 4e^{-} \rightarrow 4OH^{-}_{(aq)}$   
(C)  $Pb^{2+} + H_2O_{(l)} + \frac{1}{2} O_{2(g)} \rightarrow Pb(OH)_2 + 2e^{-}$  (D)  $2H_{2(g)} + O_{2(g)} \rightarrow 2H_2O_{(l)} + 571.7 \text{ KJ}$
  - એક અજ્ઞાત પ્રક્રિયામાં 2 મોલ પ્રક્રિયકના મોલમાંથી 1 મોલ નીપજ બનવામાં 1 કલાક લાગે છે. 4 કલાક પછીથી પ્રક્રિયકના કેટલા મોલ બાકી રહેશે ?  
(A) 0.125 મોલ (B) સૂન્ય મોલ (C) 0.225 મોલ (D) 0.5 મોલ
  - એક આલ્વીય પ્રક્રિયાને ઊંચા દબાણે અથવા ઊંચા તાપમાને કરવાથી તેનો પ્રક્રિયા ક્રમ કેટલો થાય ?  
(A) 0 (B) 2 (C) 1 (D)  $\frac{3}{2}$
  - સામાન્ય રીતે રાસાયણિક પ્રક્રિયાનો વેગ બમણો કરવા માટે તાપમાનમાં  $10^\circ$ નો વધારો કરાય છે. જો પ્રક્રિયાના તાપમાનમાં

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**WJ Hussar**

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