

Elements, Compounds, Mixtures, & Solutions

NAME _____ ROTATION _____

Part 1: Read the following information on elements, compounds, and mixtures. Fill in the blanks.

Elements:

A pure substance containing only one kind of _____.

Compounds:

The atoms are _____ combined in some way. Often (but not always), they come together to form groups of atoms called molecules.

Mixtures:

Mixtures can be uniform (called _____) and are known as solutions.

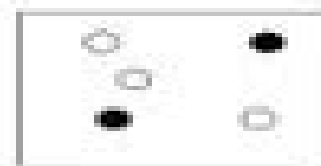
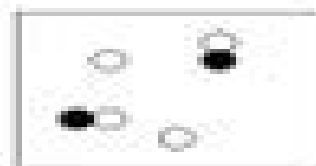
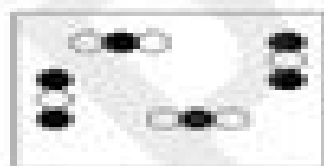
Draw a picture to represent the following:

Element	Compound	Mixture

Part 2: Classify each of the following as elements (E), compounds (C), or Mixtures (M). Write the letter X if it is none of these.

____Diamond (C)	____Sugar($C_6H_{12}O_6$)	____Milk	____Iron (Fe)
____Air	____Sulfuric Acid (H_2SO_4)	____Gasoline	____Electricity

Part 3: Match each diagram with its correct description. Diagrams will be used once.



A	B	C	D	E
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____ 1. Pure element - only one type of atom present.

____ 2. Mixture of two elements - two types of uncombined atoms present.

____ 3. Pure compound - only one type of compound present.

2 3 Elements And Compounds Section Review Answer Key

Leo J. Malone



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