

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	Very	Very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (III) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acids And Bases Section 3 Answer Key

K Morrison



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Table of Contents Acids And Bases Section 3 Answer Key

1. Understanding the eBook Acids And Bases Section 3 Answer Key
 - The Rise of Digital Reading Acids And Bases Section 3 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acids And Bases Section 3 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acids And Bases Section 3 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acids And Bases Section 3 Answer Key
 - Personalized Recommendations
 - Acids And Bases Section 3 Answer Key User Reviews and Ratings
 - Acids And Bases Section 3 Answer Key and Bestseller Lists
5. Accessing Acids And Bases Section 3 Answer Key Free and Paid eBooks
 - Acids And Bases Section 3 Answer Key Public Domain eBooks
 - Acids And Bases Section 3 Answer Key eBook Subscription Services
 - Acids And Bases Section 3 Answer Key Budget-Friendly Options

6. Navigating Acids And Bases Section 3 Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Acids And Bases Section 3 Answer Key Compatibility with Devices
 - Acids And Bases Section 3 Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acids And Bases Section 3 Answer Key
 - Highlighting and Note-Taking Acids And Bases Section 3 Answer Key
 - Interactive Elements Acids And Bases Section 3 Answer Key
8. Staying Engaged with Acids And Bases Section 3 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acids And Bases Section 3 Answer Key
9. Balancing eBooks and Physical Books Acids And Bases Section 3 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acids And Bases Section 3 Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Acids And Bases Section 3 Answer Key
 - Setting Reading Goals Acids And Bases Section 3 Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acids And Bases Section 3 Answer Key
 - Fact-Checking eBook Content of Acids And Bases Section 3 Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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