

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
  - Two nonmetal atoms bonded together
  - Two metal atoms bonded together
  - A metal and a nonmetal atom bonded together
  - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
  - Covalent
  - Both a and b
  - Neither a nor b
16. (4 pts) How many total atoms are in this molecule:  $C_6H_{12}OH$ ?
- 3
  - 4
  - 19
  - 20
  - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is  $C_3H_xCl_2$ , what must  $x$  be?
- 1
  - 2
  - 3
  - 4
  - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water?  $C_3H_6ClF$   $C_6H_5CH_3$   $CS_2$   $BaSO_4$
- $C_3H_6ClF$
  - $C_6H_5CH_3$
  - $CS_2$
  - $BaSO_4$
  - $CS_2$  and  $BaSO_4$

# Answers For Apologia Chemistry Module 14 Test

**D Keegan**



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